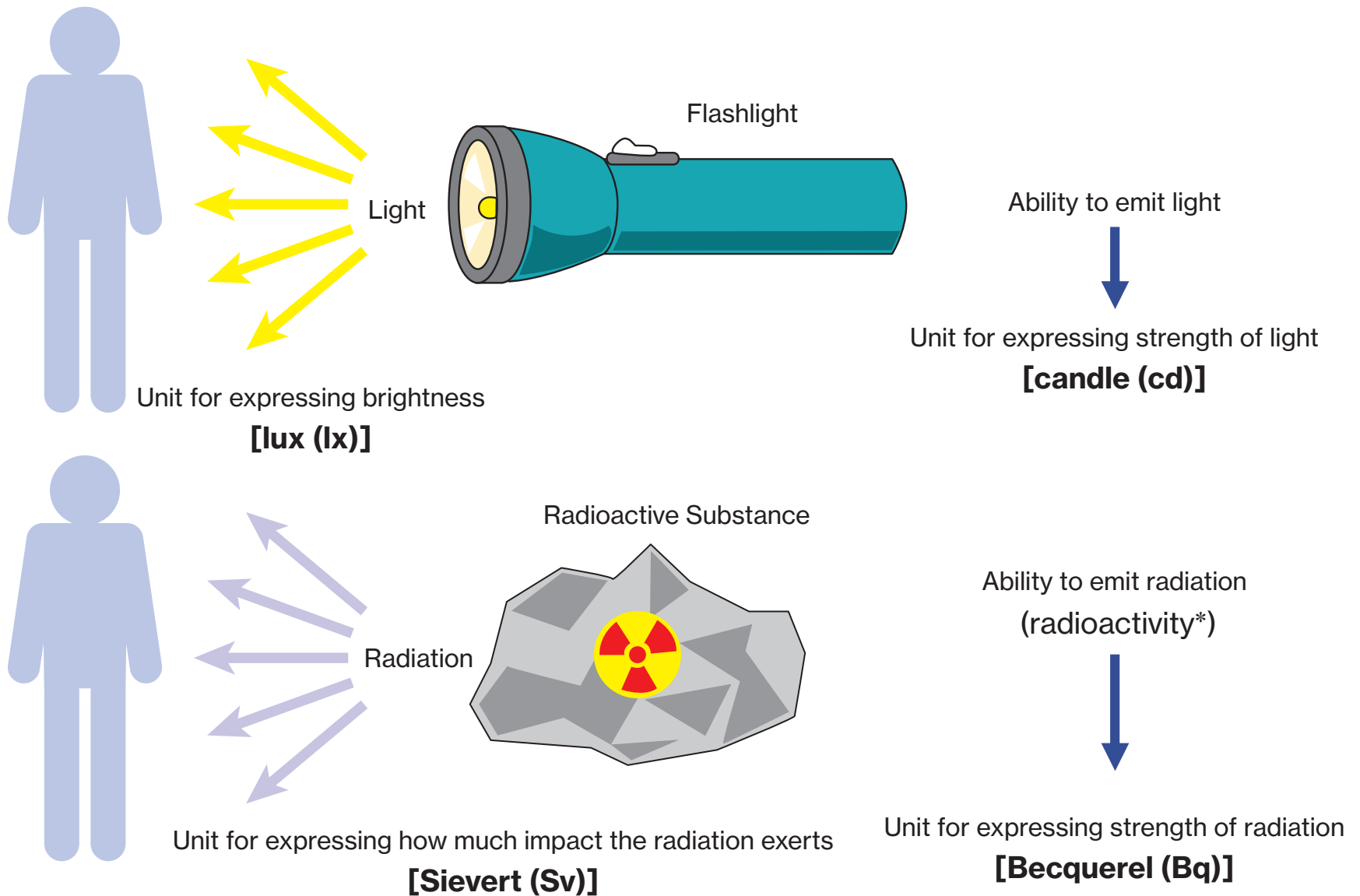


# Radioactivity and Radiation

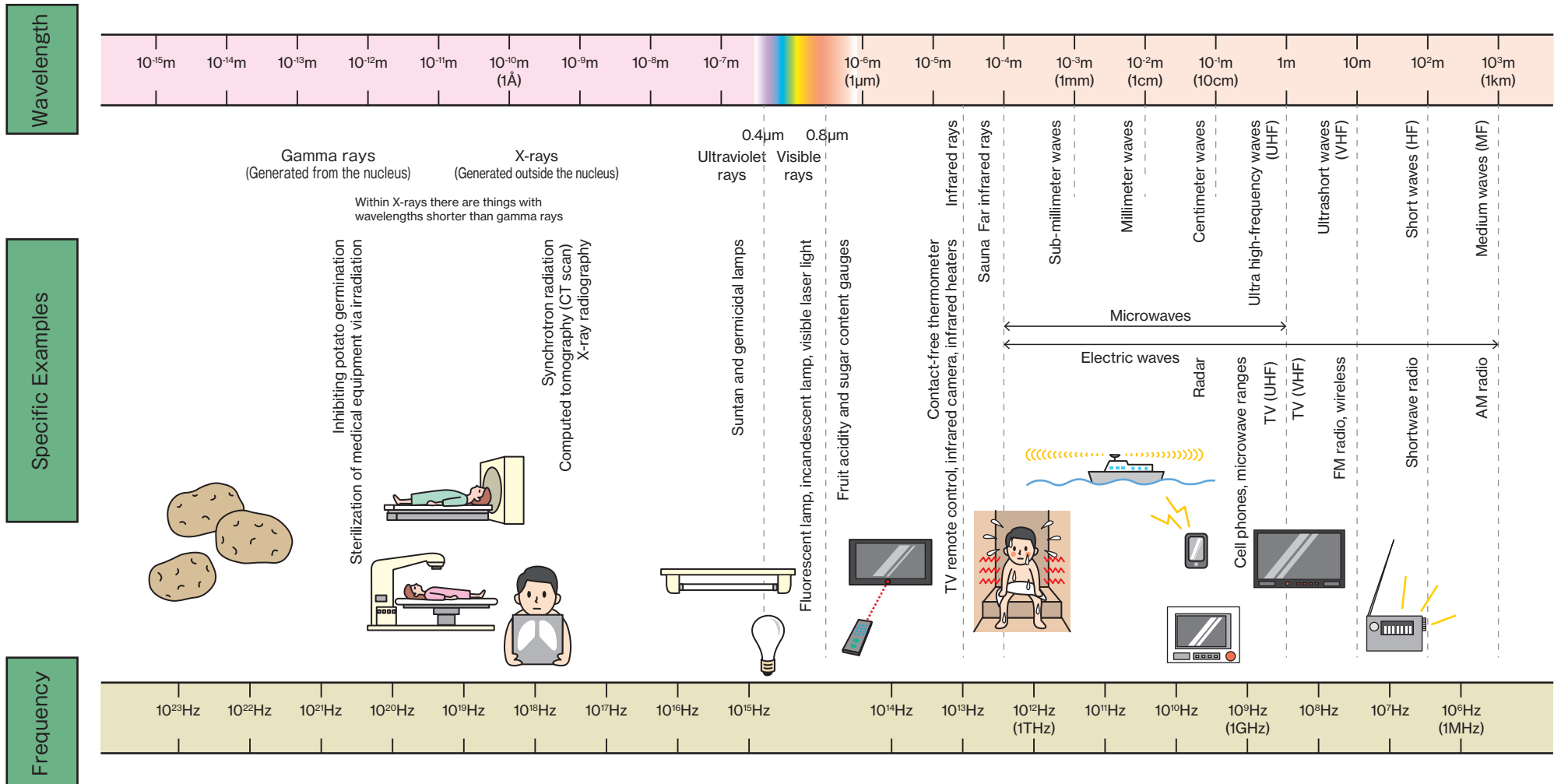


\*May also be used to describe a substance (radioactive material) that is radioactive.

# Units Used in Relation to Radioactivity

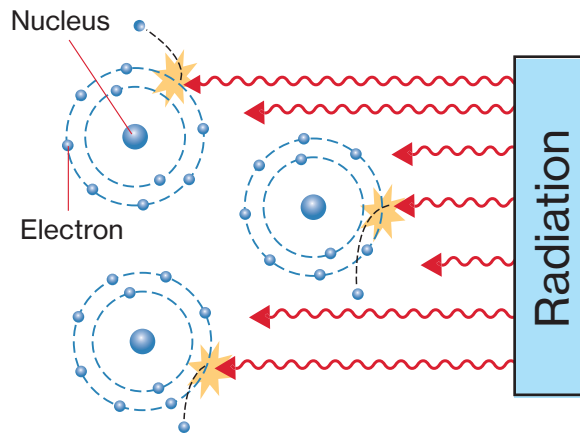
Name	Unit (Symbol)	Definition
Unit of Radioactivity International System of Units (SI)		
<b>Radioactivity</b>	Becquerel (Bq)	This unit represents the quantity of radioactive material in which one nucleus decays per second.
Unit of Radiation Dosage International System of Units (SI)		
<b>Absorbed dose</b>	Gray (Gy)	This unit represents how much energy is received by an object or person hit by radiation. A dose of 1 Gray corresponds to 1 joule of energy absorbed by 1 kilogram of matter.
<b>Dose</b>	Sievert (Sv)	This unit is used for assessing how much risk radiation poses to people in terms of inducing cancer or genetic damage. (1 Sievert = 1,000 mSv)
Unit of Energy International System of Units (SI)		
<b>Energy</b>	Joule (J)	Unit of energy that represents the energy of radiation (1J=6.2×10 <sup>18</sup> eV)

# Types of Electromagnetic Waves

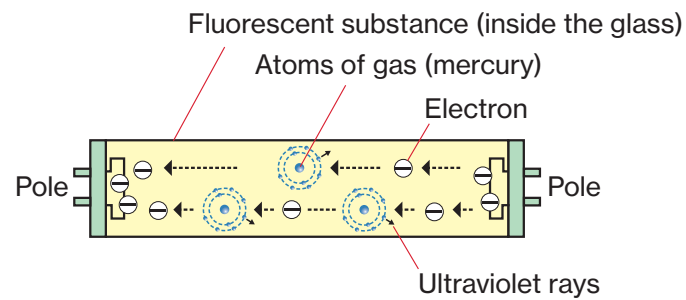


# Properties of Radiation

## Ionization Effect



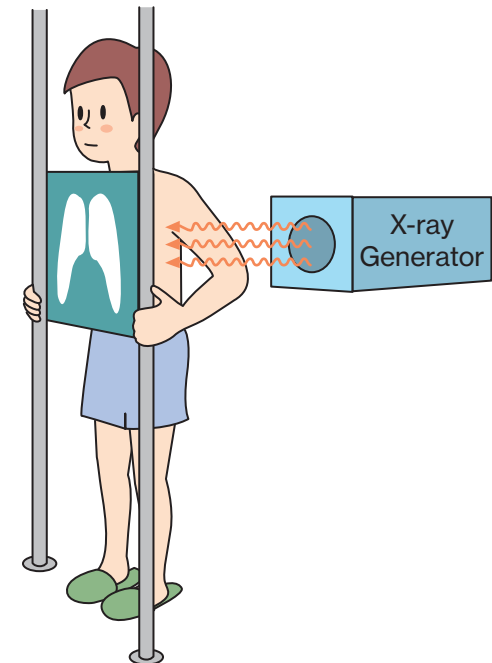
## Fluorescence Effect



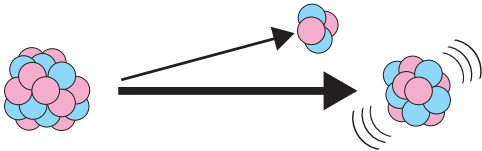
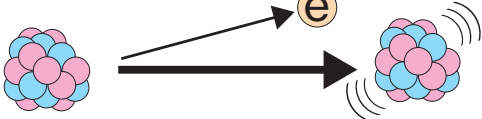
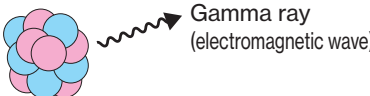
### How Fluorescence Works

When voltage is applied across the poles of the tube, electrons flow from one pole to the other. When electrons crash into mercury enclosed in the tube, it produces ultraviolet light. The ultraviolet rays light up the fluorescent substance.

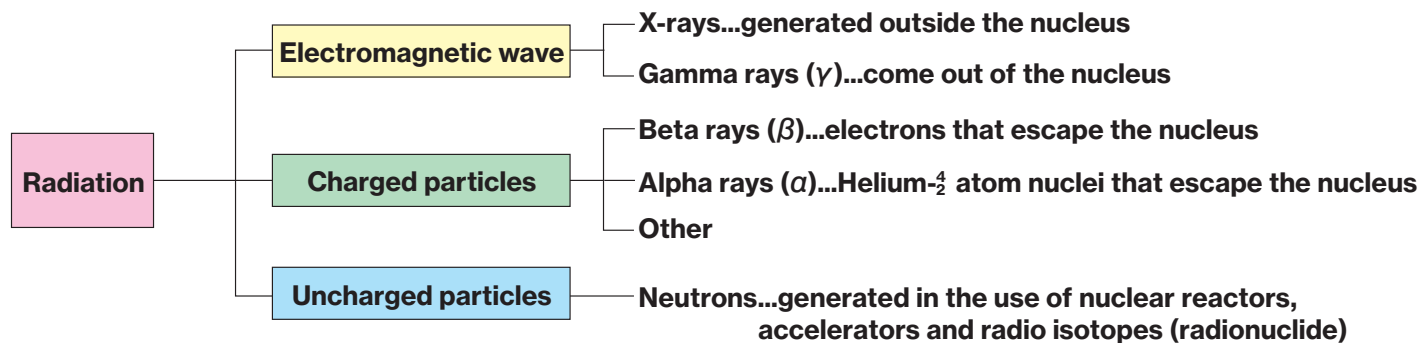
## Permeation



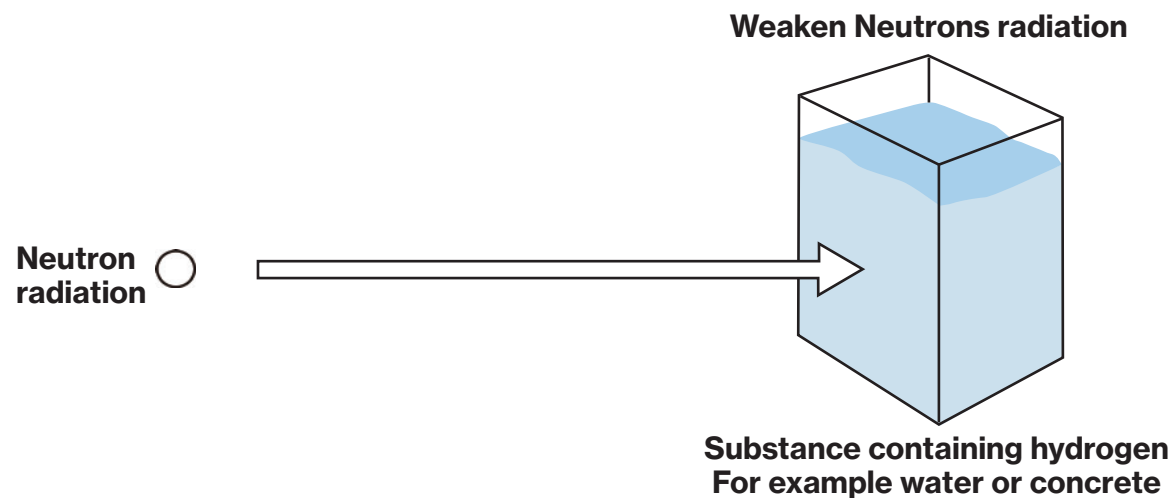
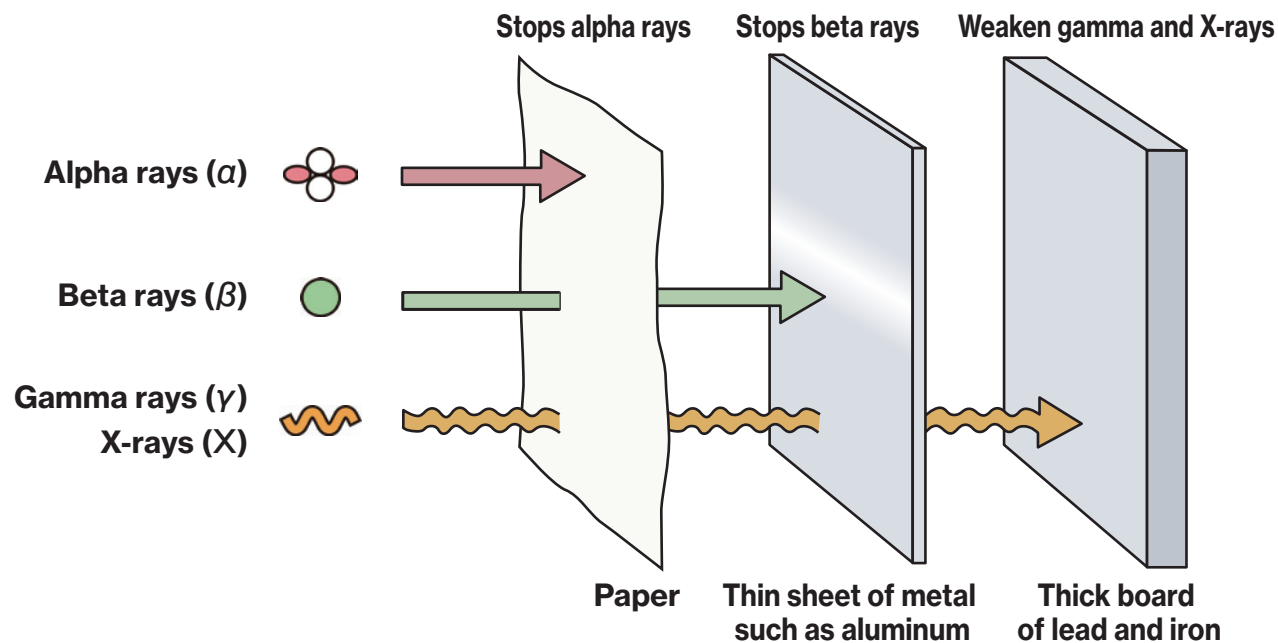
# Types of Radiation

<b>Alpha (<math>\alpha</math>) decay</b>	<p>Alpha ray (<math>\frac{4}{2}\text{He}</math> nucleus)</p> 	<p>(Ex.)</p> ${}_{88}^{226}\text{Ra} \xrightarrow{\alpha} {}_{86}^{222}\text{Rn}$
<b>Beta (<math>\beta</math>) decay</b>	<p>Beta ray (electron)</p> 	<p>(Ex.)</p> ${}_{11}^{24}\text{Na} \xrightarrow{\beta} {}_{12}^{24}\text{Mg}$
<b>Gamma (<math>\gamma</math>) ray emission</b>	 <p>Gamma ray (electromagnetic wave)</p>	

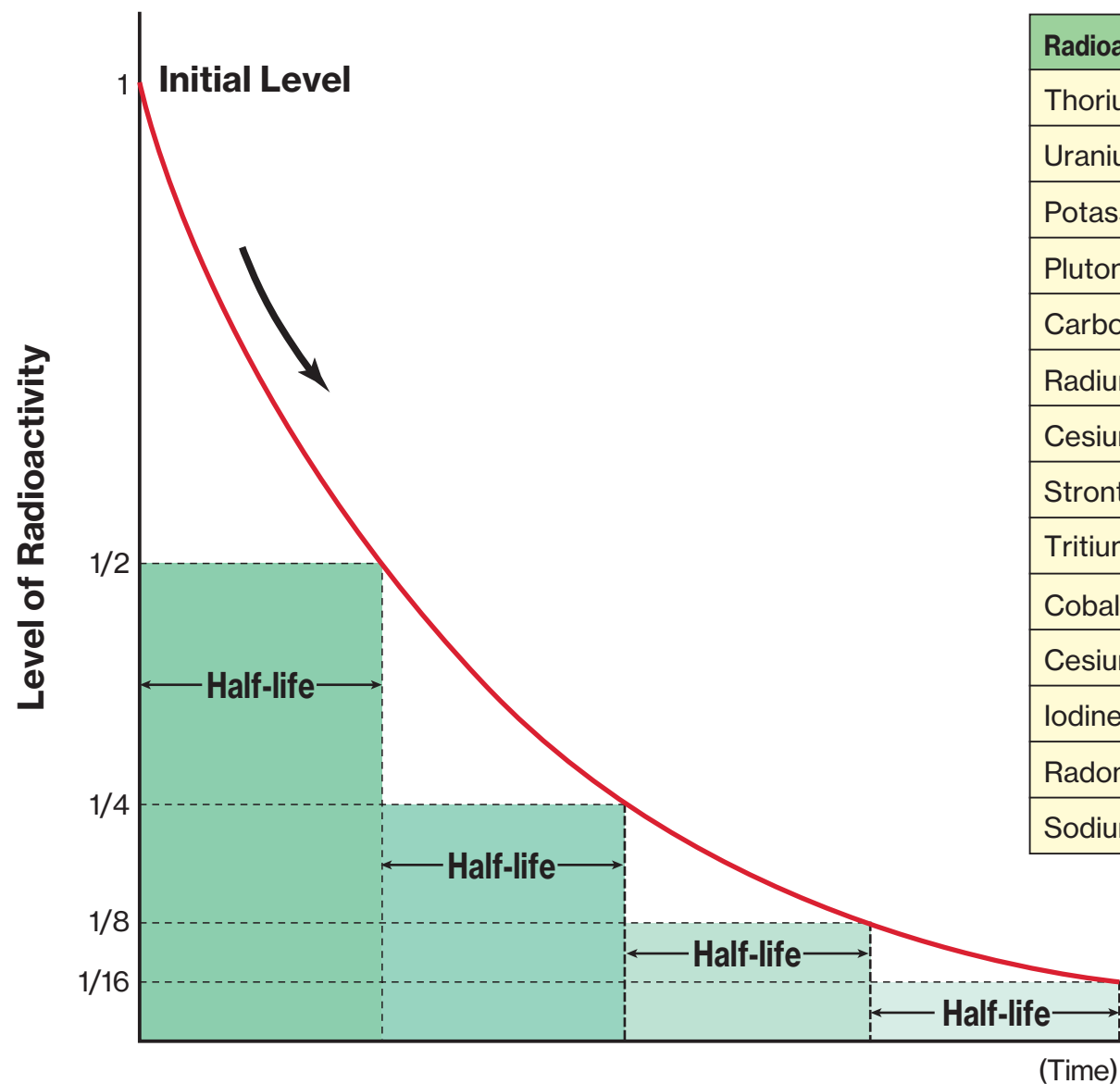
● Proton ● Neutron



# Permeability of Different Kinds of Radiation



# How Radiation Decays



Radioactive Substance	Emitted Radiation*	Half-life
Thorium-232	$\alpha \cdot \beta \cdot \gamma$	14.1 billion years
Uranium-238	$\alpha \cdot \beta \cdot \gamma$	4.5 billion years
Potassium-40	$\beta \cdot \gamma$	1.3 billion years
Plutonium-239	$\alpha \cdot \gamma$	24,000 years
Carbon-14	$\beta$	5,700 years
Radium-226	$\alpha \cdot \gamma$	1,600 years
Cesium-137	$\beta \cdot \gamma$	30 years
Strontium-90	$\beta$	28.8 years
Tritium	$\beta$	12.3 years
Cobalt-60	$\beta \cdot \gamma$	5.3 years
Cesium-134	$\beta \cdot \gamma$	2.1 years
Iodine-131	$\beta \cdot \gamma$	8 days
Radon-222	$\alpha \cdot \gamma$	3.8 days
Sodium-24	$\beta \cdot \gamma$	15 hours

\*Includes radiation from products of decay (Nuclides that emit radiation and become a different nuclide.)

# Radiation in our Daily Lives



We receive doses of radiation in our daily lives from a variety of sources.



From radon in the air\*2  
0.47



From food 0.99



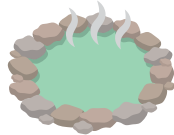
From the ground 0.33



From outer space 0.3



Aircraft use 0.008



Hot spring underground environment 0.005

Approx. 6000 mGy Permanent infertility (Testicular)\*1  
Approx. 3000 mGy Permanent infertility (Ovaries)\*1

Approx. 4000 mGy Temporary hair loss\*1

## 100 mSv or less

Amount of radiation (mSv) exposure that bears no statistical difference in terms of the risk of cancer

0.5 to 613.2 Natural radiation from the ground, Ramsar (Iran), Kerala, Chennai (India)

(No adverse impact to the health of residents found.)

**2.4 Natural radiation/person (per year)**  
(World Average)

**2.1 Natural radiation/person (per year)**  
(Average in Japan)

0.01 (year) Clearance level\*3

0.008 Aircraft use



Less than 0.001 Records of radioactive material released from nuclear power plants

## Radiation Exposure (mSv)

10000

1000

100

10

1

0.1

0.01

0.001

100 to 6,200 mGy  
Cardiac catheterization (skin)

500 mGy  
Cataract (vision impairment)\*1

500 mGy  
Reduced blood-forming function (bone marrow)\*1

250 Dose limits for emergency workers at power plants, etc.\*5

50 Dose limits for workers at power plants, etc. (per year)\*4

5 to 30 CT (1 time)

2 to 20 PET scan (1 time)

3.0 Stomach X-ray (1 time)

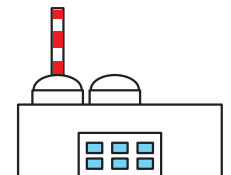
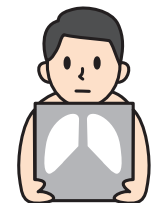
1.0 Dose limit for the general public (excluding for medical treatment) (per year)

0.06 Chest X-ray in annual checkup (1 time)

0.05 Targeted dose around a nuclear plant (per year)

0.022 Estimated dose at reprocessing plant (Rokkasho) (per year)

0.002 to 0.01 Dental X-ray



\*1:When discussing radiation hazards, it is expressed as equivalent to an effective dose of 1 mSv, given that a dose of 1 mSv of gamma radiation is absorbed evenly by each part of the the entire body

\*2:Radioactive substances naturally present in the air

\*3:Insignificant compared to naturally-occurring radiation levels, and the level does not require handling as a radioactive substance that presents a safety risk.

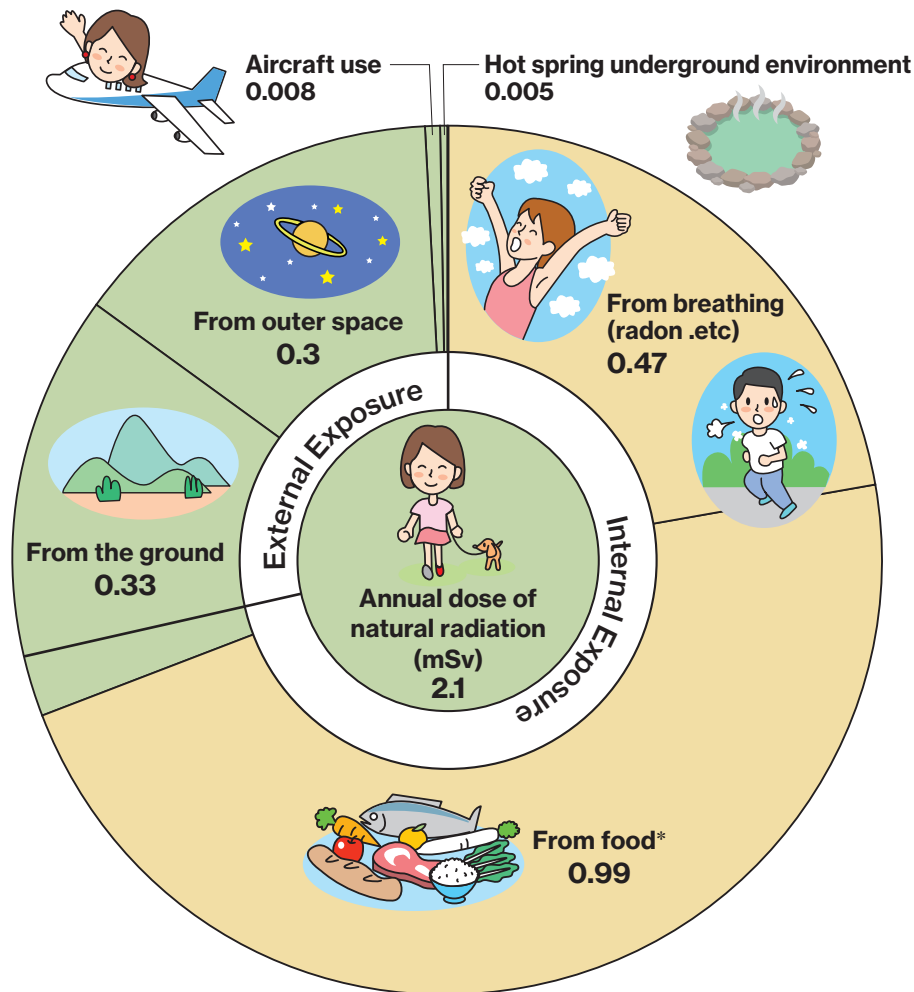
\*4:Dose of radiation that must not be exceeded in 1 year is 50 mSv for workers at places such as power stations, or 100 mSv over 5 years.

\*5:The dose limit was raised to 250 mSv to emergency workers from April 2016 due to the revision of the Ionizing Radiation Hazard Prevention Regulations, etc.

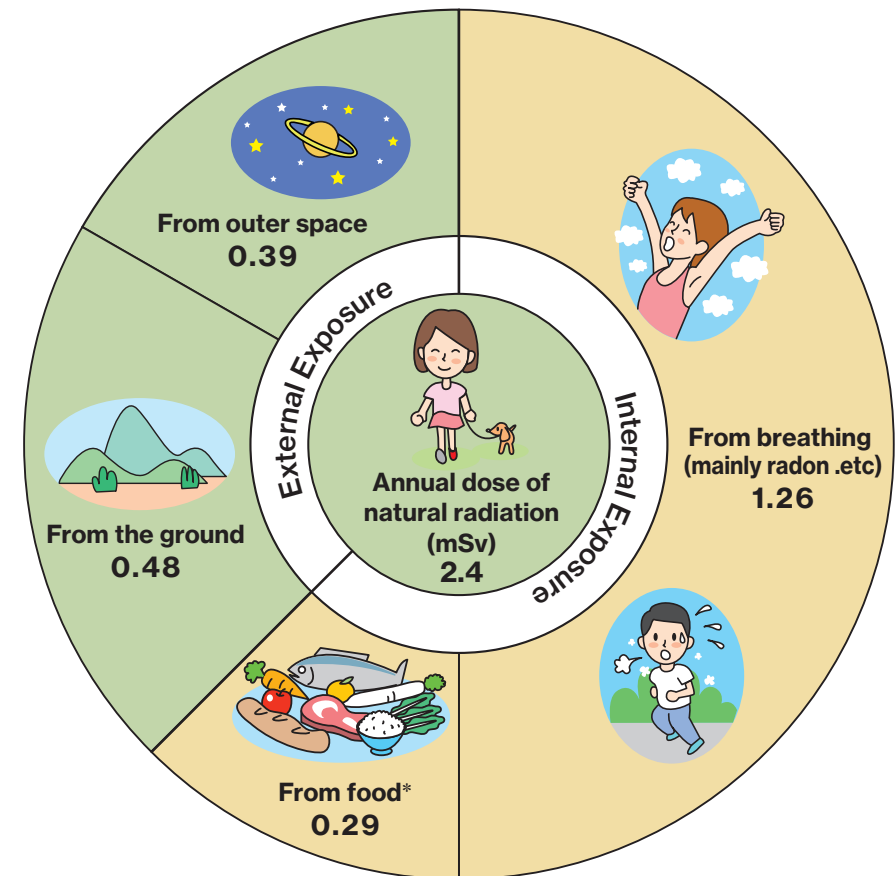


# Doses of Radiation from Natural Sources

Annual dose/person (average in Japan)

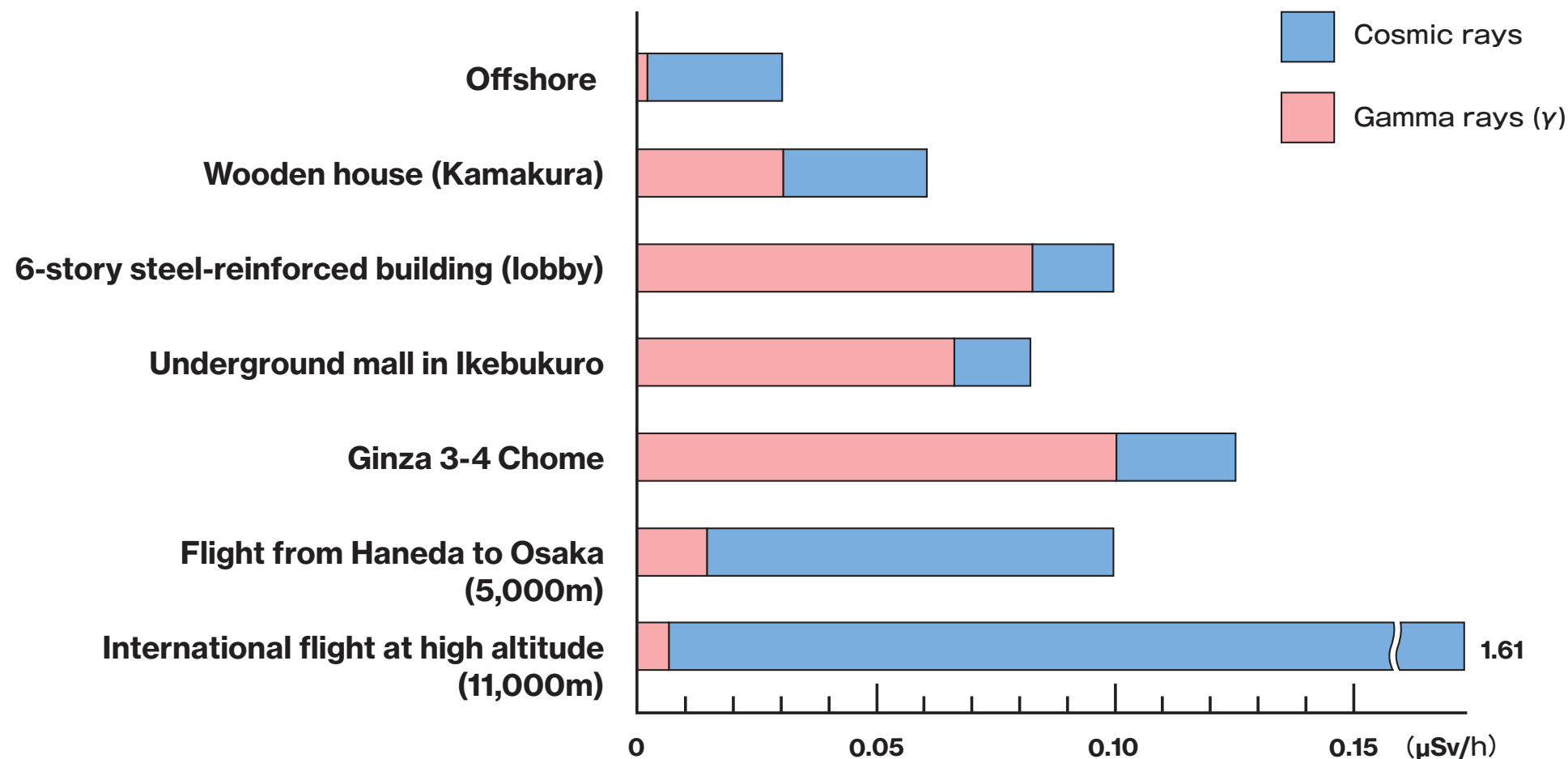


Annual dose/person (average worldwide)



\*Compared to Western countries, the Japanese diet of seafood results in a larger effective dose due to Polonium-210.

# Differences in Natural Radiation Levels



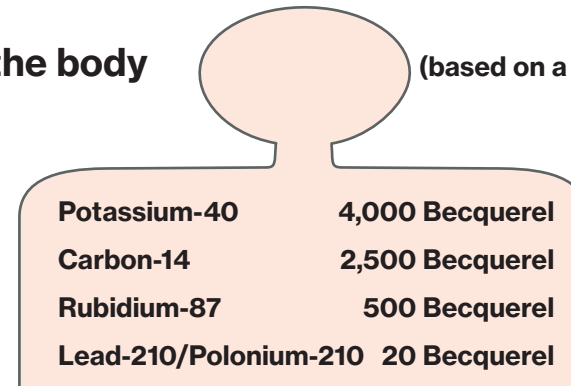
\* $1\mu\text{Sv}=1/1000\text{ mSv}$

$1\mu\text{Sv/h}=365\text{ days} \times 24\text{hours} \times 1,000=8.75\text{mSv/year}$

# Naturally Occurring Radiation in the Body & Our Food

## ●Level of radioactive materials in the body



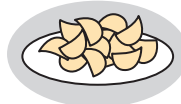
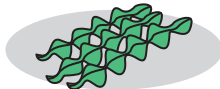




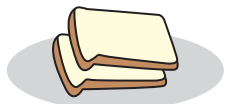


(based on a Japanese person weighing 60kg)



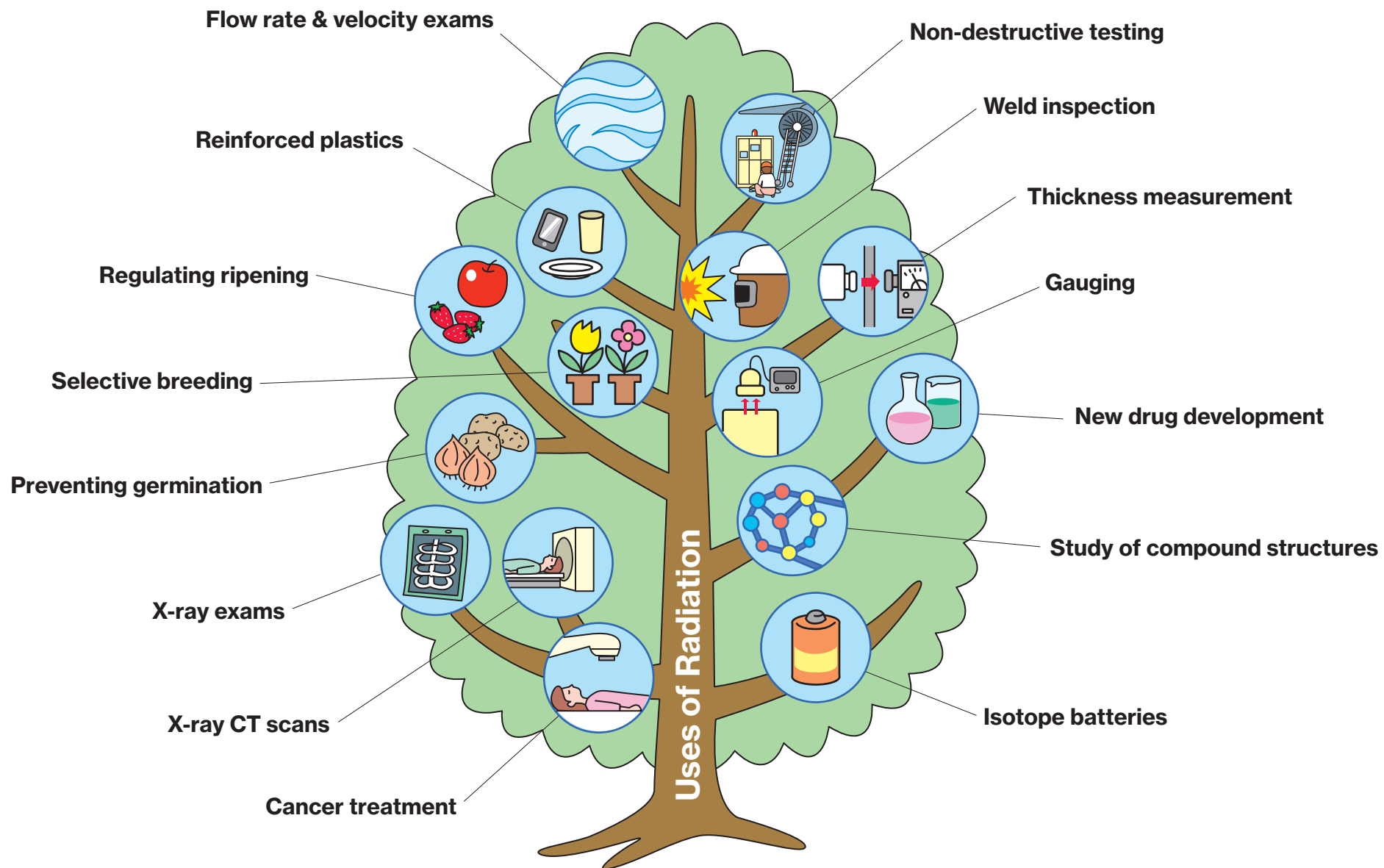
Potassium-40	4,000 Becquerel
Carbon-14	2,500 Becquerel
Rubidium-87	500 Becquerel
Lead-210/Polonium-210	20 Becquerel

## ●Level of the radioactive material Potassium-40 in food (Japan)


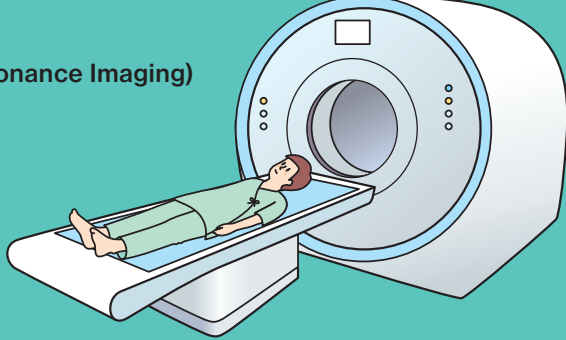
(Unit: Becquerel/kg)

			
Dried kelp 2,000	Dried shiitake 700	Potato chips 400	
			
Raw seaweed 200	Spinach 200	Fish 100	Beef 100
			
Milk 50	Bread 30	Rice 30	Beer 10

# Various Uses of Radiation



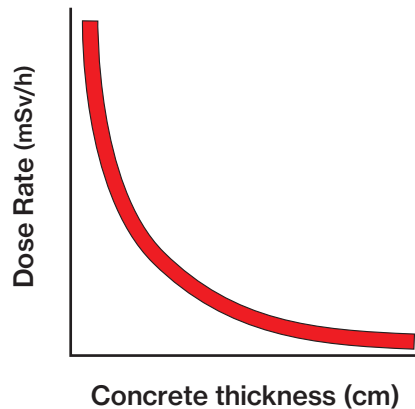
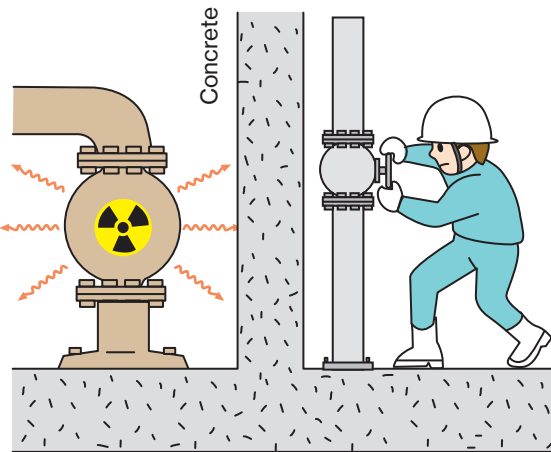
# Medical Exams: Differences Between X-Ray CT and MRI

	<b>X-Ray CT</b> (Computed Tomography) 	<b>MRI</b> (Magnetic Resonance Imaging) 
<b>Mechanism</b>	Uses radiation (X-rays)	Uses magnetism and electromagnetic waves
<b>Time required and effect of patient movement</b>	Short examination time (1 body part: about 10 min.)	Long examination time (1 body part: about 20 to 60 min.), image becomes distorted easily if the patient moves
<b>Noise</b>	Quiet	Loud
<b>Radiation exposure</b>	Yes	No
<b>Tissue contrast</b>	Inferior to MRI for some body parts	Very clear contrast
<b>Detailed rendering</b>	Possible (better than MRI)	Possible
<b>Body parts enclosed in bone</b>	Highly distorted image	Low distortion of image
<b>Imaging of bleeding condition</b>	Possible	Possible (better than X-ray CT)
<b>Imaging of condition of blood vessels</b>	Not possible (but possible by using radiocontrast agent)	Possible
<b>If metal is inside body</b>	Imaging is possible (but may not be possible if there is a pacemaker or other device)	Imaging not possible (but may be possible if the metal inside body is non-reactive to magnets)

# Basics of Protection from Radiation

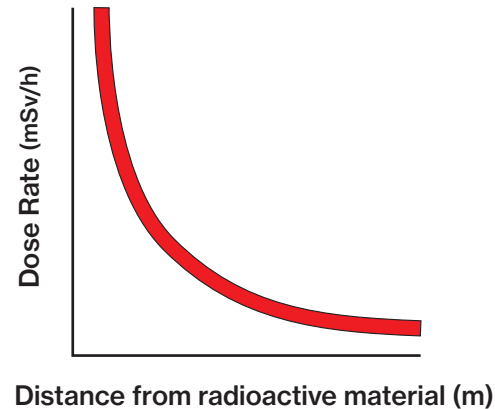
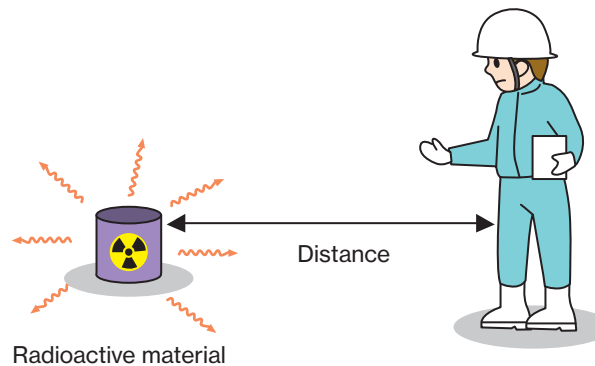
## 1. Protection via shielding

Dose Rate = drops as the shield becomes thicker



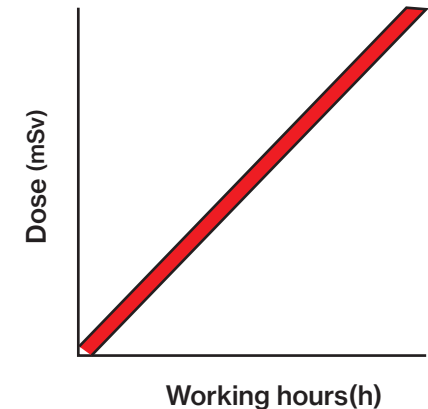
## 2. Protection via distance

Dose Rate = inversely proportional to the distance

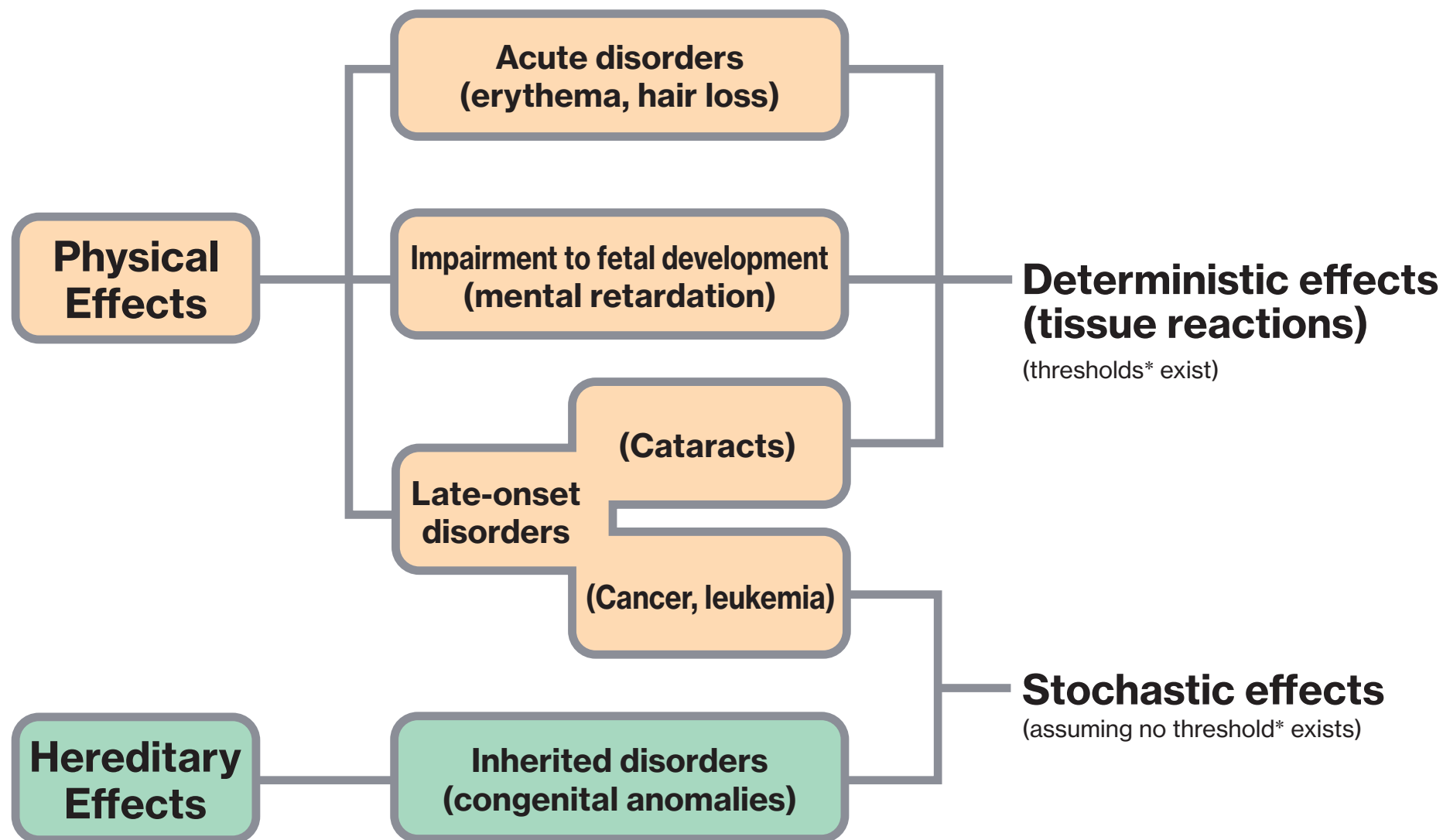


## 3. Protection via time

Dose = (dose rate in work zone) X (working hours)

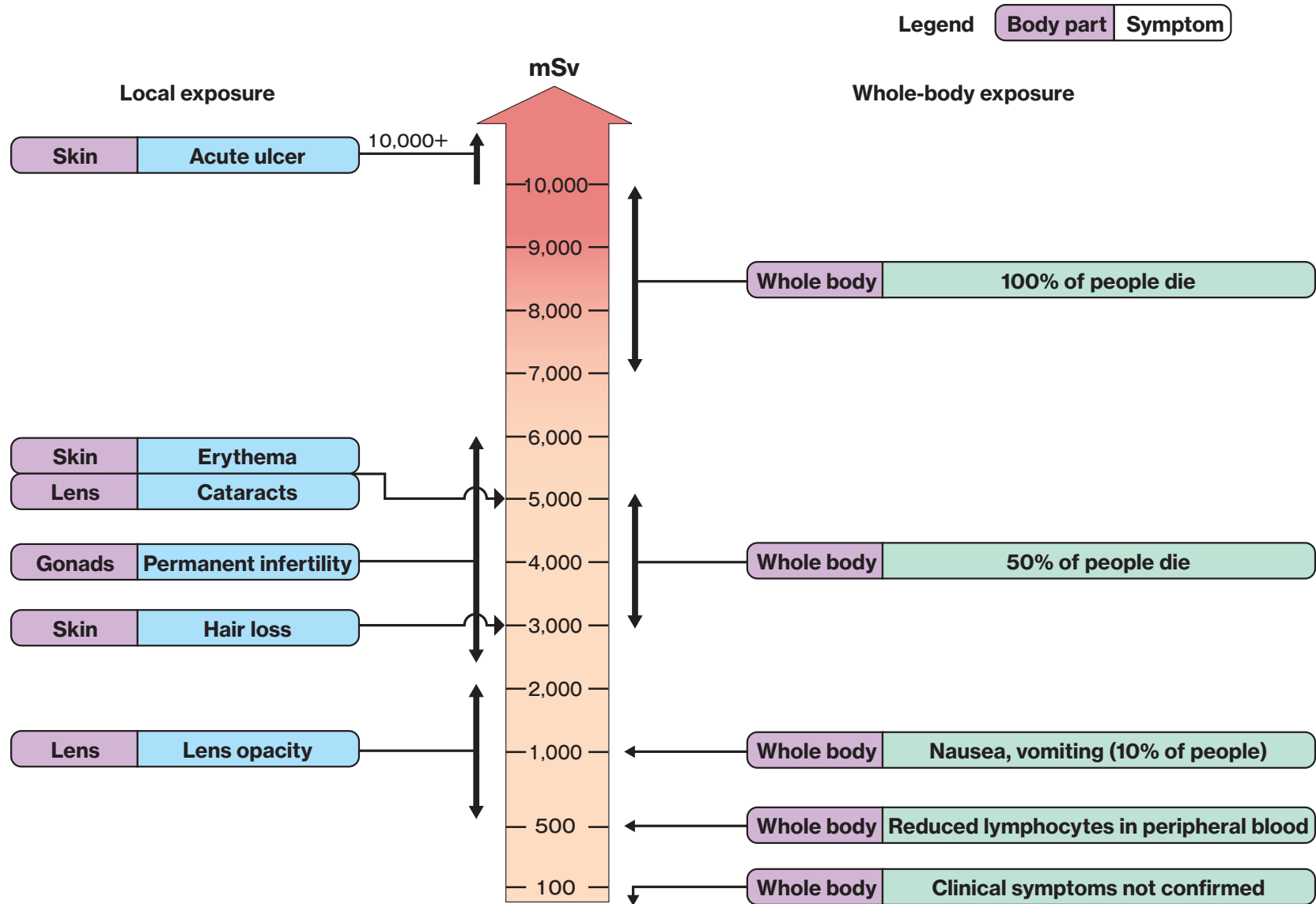


# Effects of Radiation on Human Health



\*Threshold: value that constitutes a turning point at which a reaction occurs or not.

# Symptoms from a Single Exposure to Radiation



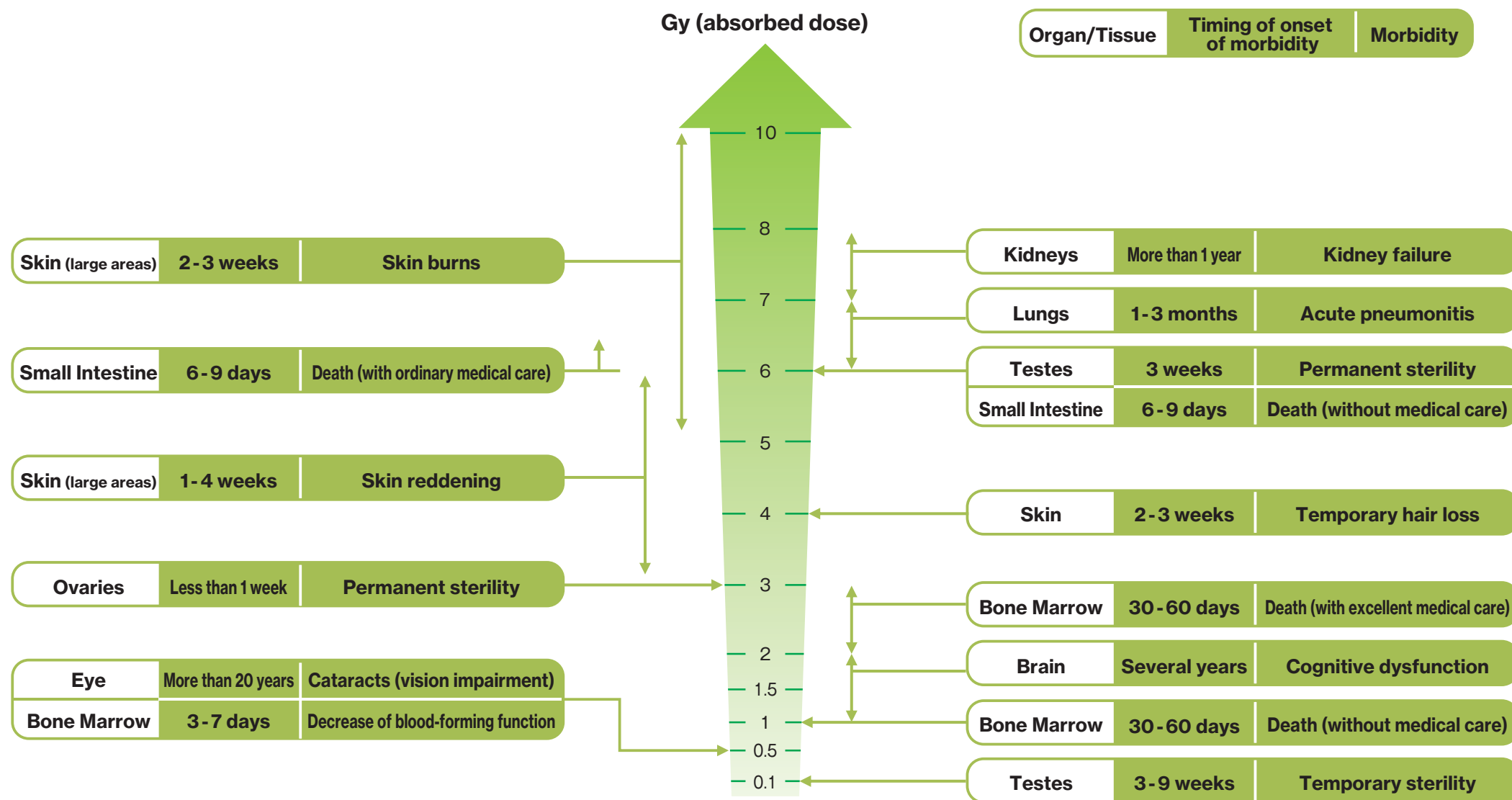
(Note 1) Deterministic effects (tissue reaction) are described, except for cancer and hereditary effects.

(Note 2) General public dose limit is 1.0mSV/year, dose limit around a nuclear plant is 0.05mSV/year.



# Acute Radiation Effects

Threshold for 1% incidence of major morbidity and death after acute exposure to gamma rays\*

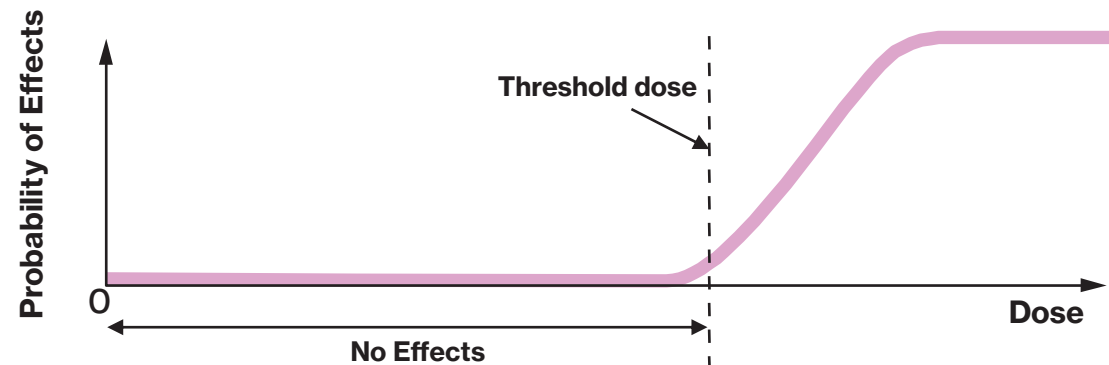


\*Threshold: The maximum level of radiation considered to be acceptable or safe

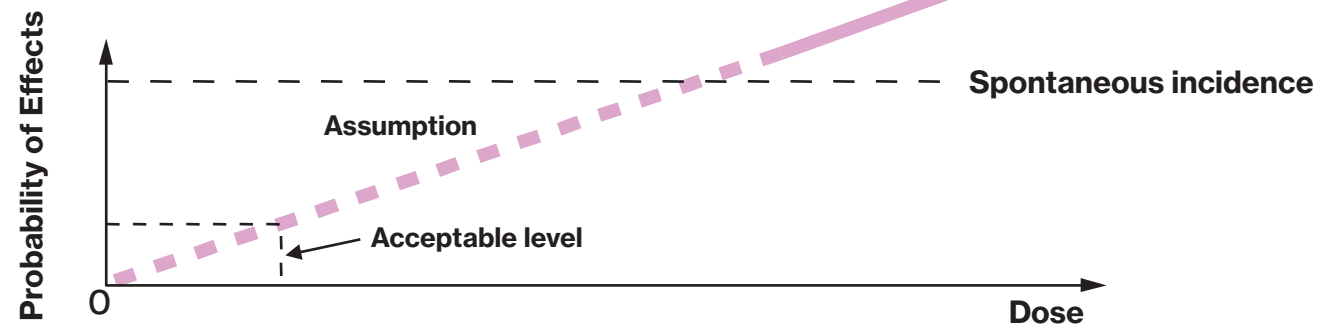
# Principles of Protection from Radiation

Below the threshold dose\* for deterministic effects, the effect of radiation is suppressed and/or eliminated. For stochastic effects, the assumption is that there is no threshold dose below which radiation can be suppressed and be relatively certain that an adverse effect cannot occur.

[Deterministic effects (tissue reactions): hair loss, cataracts, etc.]



[Stochastic effects: cancer, leukemia, etc.]

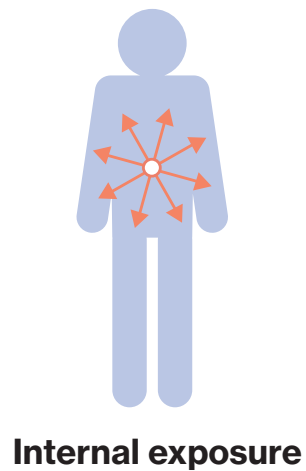
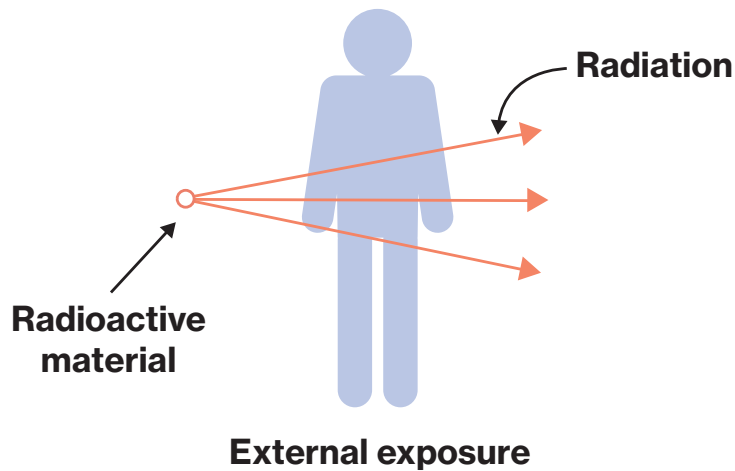


\*Threshold dose: value that constitutes a turning point at which a reaction occurs or not.

# The Difference between Exposure and Contamination

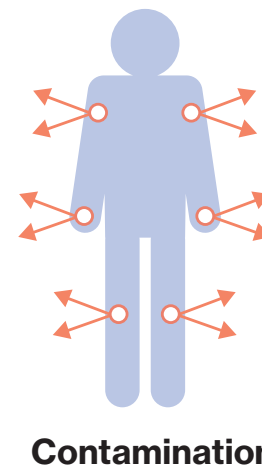
## Exposure

Being exposed to radiation



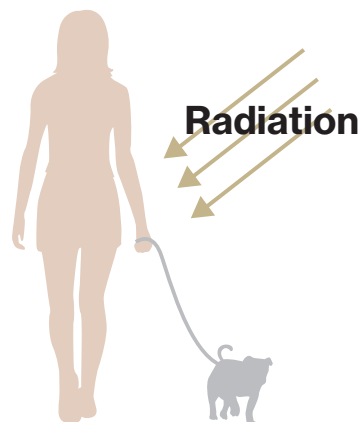
## Contamination

Radioactive materials have adhered to matter, such as skin or clothes.



# The Relationship between Gray & Sievert Units

$$\text{Sievert value} = \text{Gray value} \times \text{Radiation weighting factor}^{*1} \times \text{Tissue weighting factor}^{*2}$$



## Sievert (Sv)

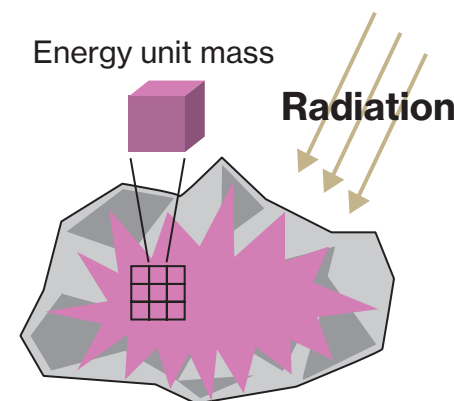
This unit is used for assessing how much risk radiation poses to people in terms of inducing cancer or genetic damage.  
(1 Sievert = 1,000 mSv)

### ◆ Radiation weighting factor

Types of Radiation	Radiation weighting factor
Photon (gamma or x-rays)	1
Electrons (beta rays)	1
Proton	2
Alpha particles, fission fragments, heavy nuclei	20
Neutron radiation	2.5-20 (Determined by the continuous function of energy)

\*1: Represents the difference in effect according to the type of radiation.

\*2: Represents how susceptible different tissues, such as internal organs, are to radiation.



## Gray (Gy)

This unit represents how much energy was received by an object or person hit by radiation.  
A dose of 1 gray corresponds to 1 joule of energy absorbed by 1 kilogram of matter.

### ◆ Tissue weighting factor

Tissue/Organ	Tissue weighting factor	Tissue/Organ	Tissue weighting factor
Breast	0.12	Esophagus	0.04
Red bone marrow	0.12	Thyroid	0.04
Colon	0.12	Salivary gland	0.01
Lung	0.12	Skin	0.01
Stomach	0.12	Bone surface	0.01
Gonads	0.08	Brain	0.01
Bladder	0.04	Remaining tissues /organs	0.12
Liver	0.04		

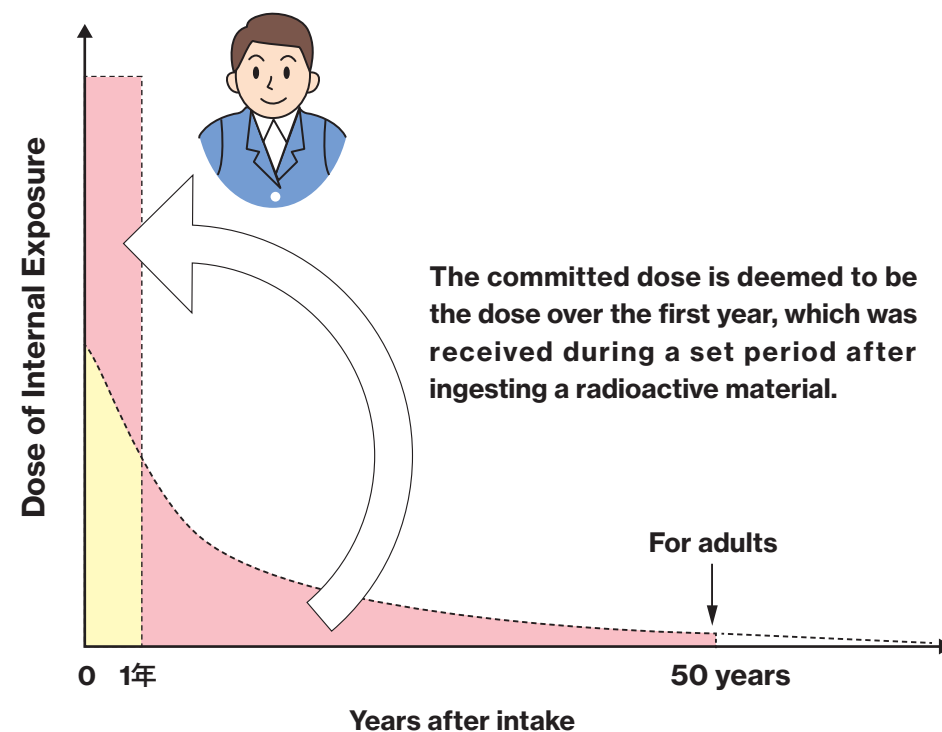
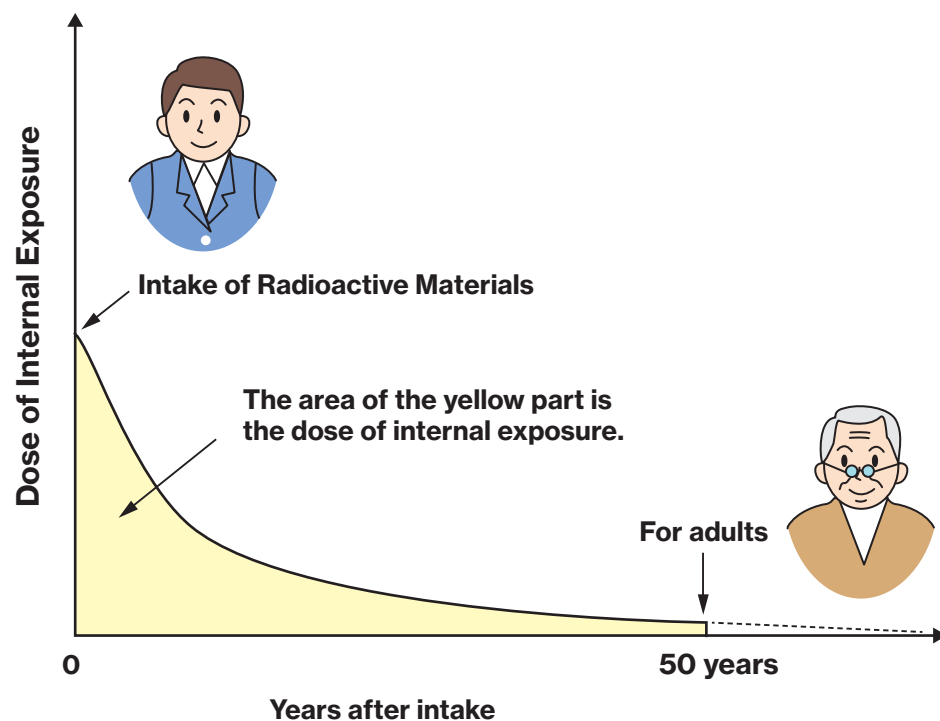
# Conversion Method for Internal Exposure to Radiation (Committed Dose)

$$\text{Committed Dose (mSv)} = \text{Food Intake (kg/day)} \times \text{Intake Days (days)} \times \text{Effective Dose Coefficient (mSV/Bq)} \times \text{Concentration of Radionuclides (Bq/kg)}$$

Radionuclide	Half-life	Effective dose coefficient when 1Bq is taken orally or inhaled by an adult (mSv/Bq)	
		Taken orally	Inhaled
Plutonium-239	24,000 years	$2.5 \times 10^{-4}$	$1.2 \times 10^{-1}$
Cesium-137	30 years	$1.3 \times 10^{-5}$	$3.9 \times 10^{-5}$
Iodine-131	8 days	$2.2 \times 10^{-5}$	$7.4 \times 10^{-6}$
Strontium-90	29.1 years	$2.8 \times 10^{-5}$	$1.6 \times 10^{-4}$
Tritium*	12.3 years	$4.2 \times 10^{-8}$	$2.6 \times 10^{-7}$

\* The effective dose coefficient of tritium shows OBT (organically bound tritium), which is important for dosimetric evaluation, because it is easily absorbed by living bodies and has a long biological half-life.  
 (Note) 1 is used for the market dilution factor (percentage of contaminated food intake relative to food intake of the evaluated subject) or for correction downwards, such as due to cooking.  
 When more than one value is indicated for the nuclide of a chemical form, the largest effective dose coefficient is shown.

# Evaluation of Internal Exposure to Radiation (Conceptual Diagram of Committed Dose)



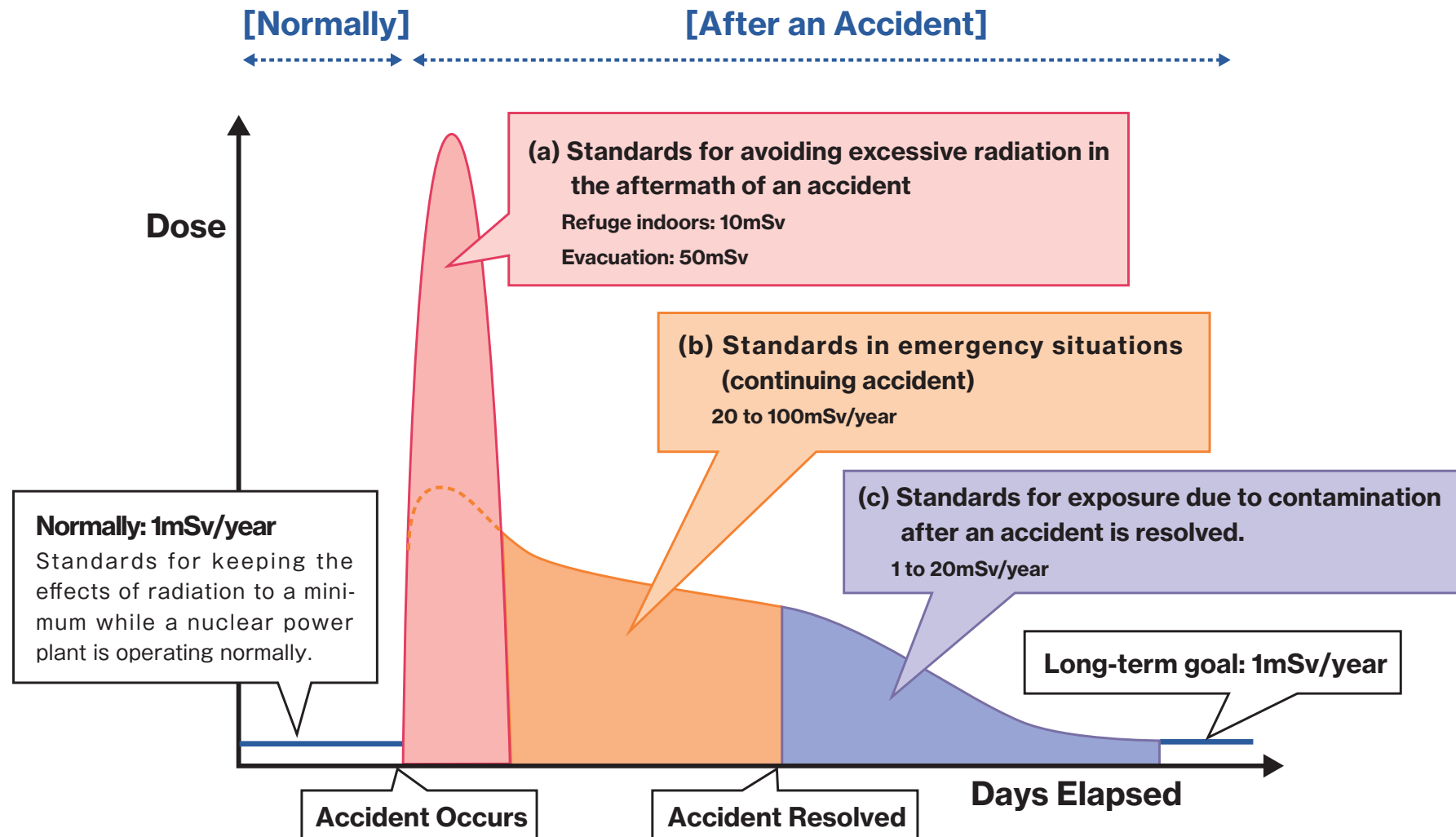
(Note) Adults : 50 years; children: from ingestion until 70 years old

# An International Comparison of Food Reference Values

(Unit: Becquerel/kg)

Nuclide	Country	Japan	U.S.A.	EU
	Food group			
Radioactive cesium	Baby food	50	1,200	400
	Milk	50		1,000
	Drinking water	10		1,000
	Foods in general	100		1,250
Principles of Food Reference Values		Established dose of radiation will be less than 1 mSv per year. Values are calculated under the assumption contamination for foods in general will be 50%; for drinking water and milk as well as baby foods, it will be 100%.	Established dose of radiation will be less than 5 mSv per year. Values are calculated under the assumption contamination for foods in general will be 30%.	Established dose of radiation will be less than 1 mSv per year. Values are calculated under the assumption contamination for foods in general will be 10%.

# Standard Principles for Protection from Radiation



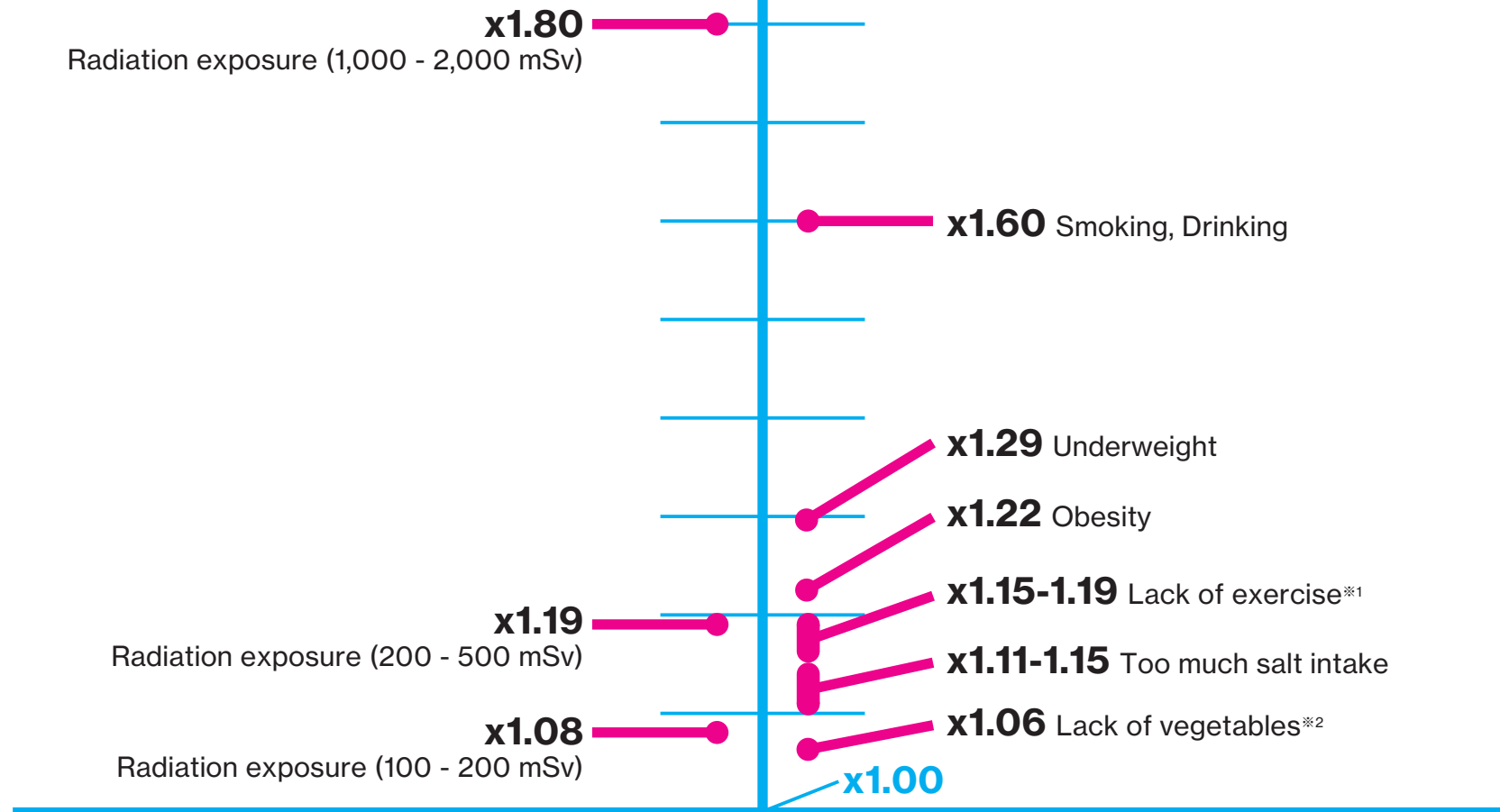


# Relative Cancer Risk Estimation for Radiation Exposure and Lifestyle Factors

(Targeted for Japanese between the ages of 40 and 69)

## Cancer Risk Factors from Exposure to Radiation

## Cancer Risk Factors related to Lifestyle

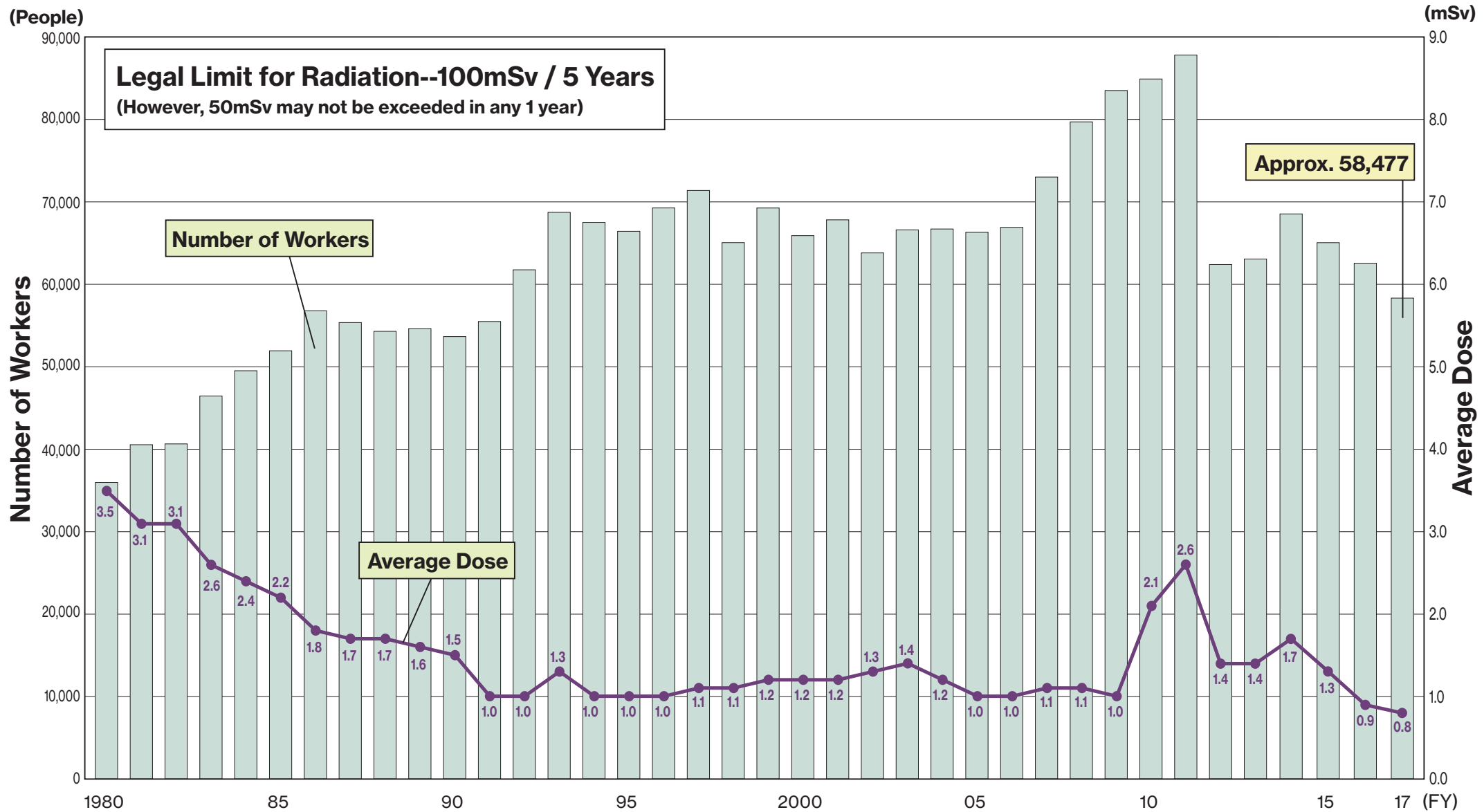


(Note) Risk estimates for radiation exposure are based on the analysis of instantaneous dose effects in the case of A-bomb at Hiroshima and Nagasaki (solid cancer incidence only) and not based on the observations on prolonged dose effects.

※1 Lack of exercise: a very small amount of physical activity

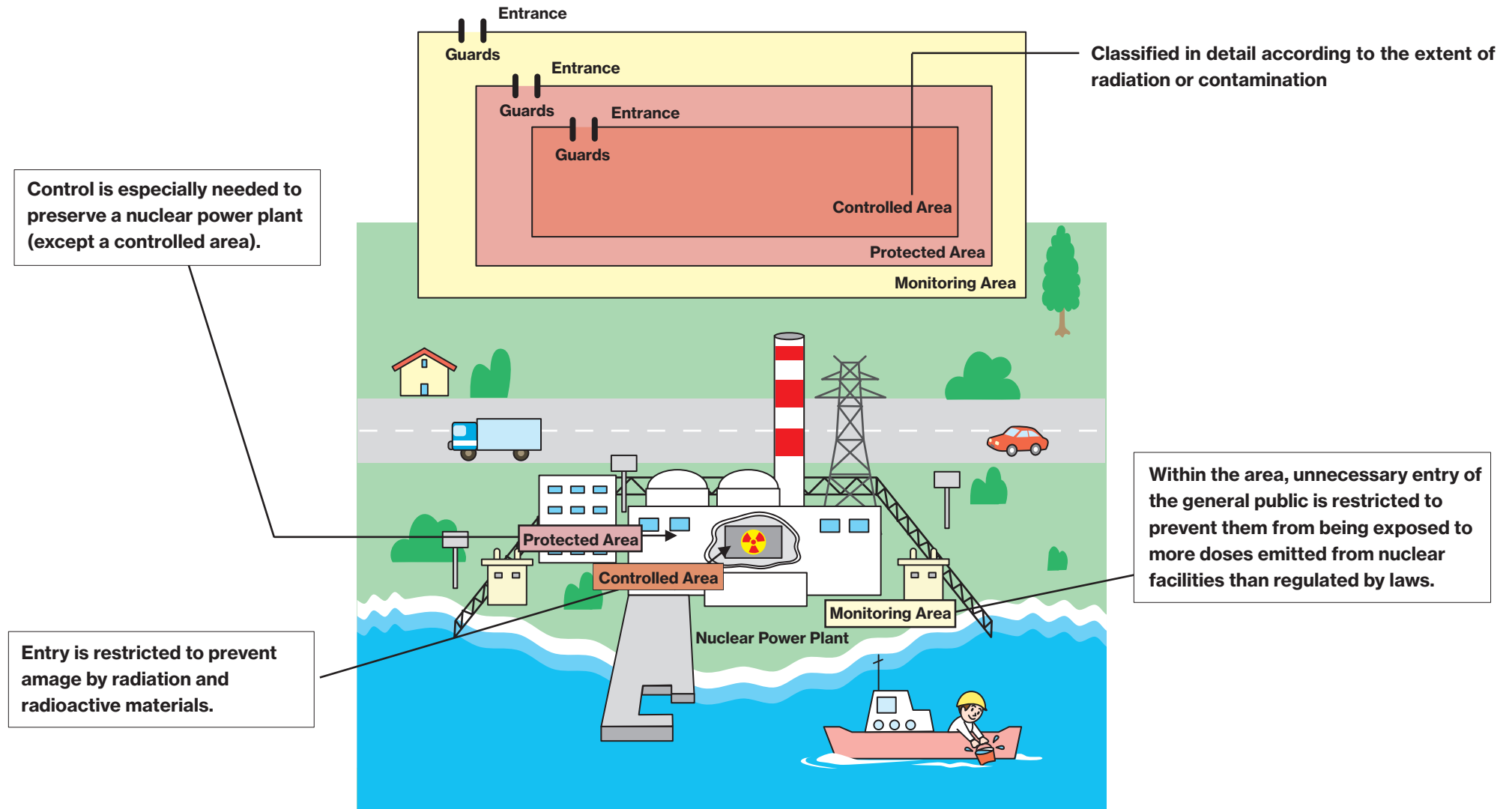
※2 Lack of vegetables: vegetable intake is very low

# Radiation Dosages Received by Radiation Workers



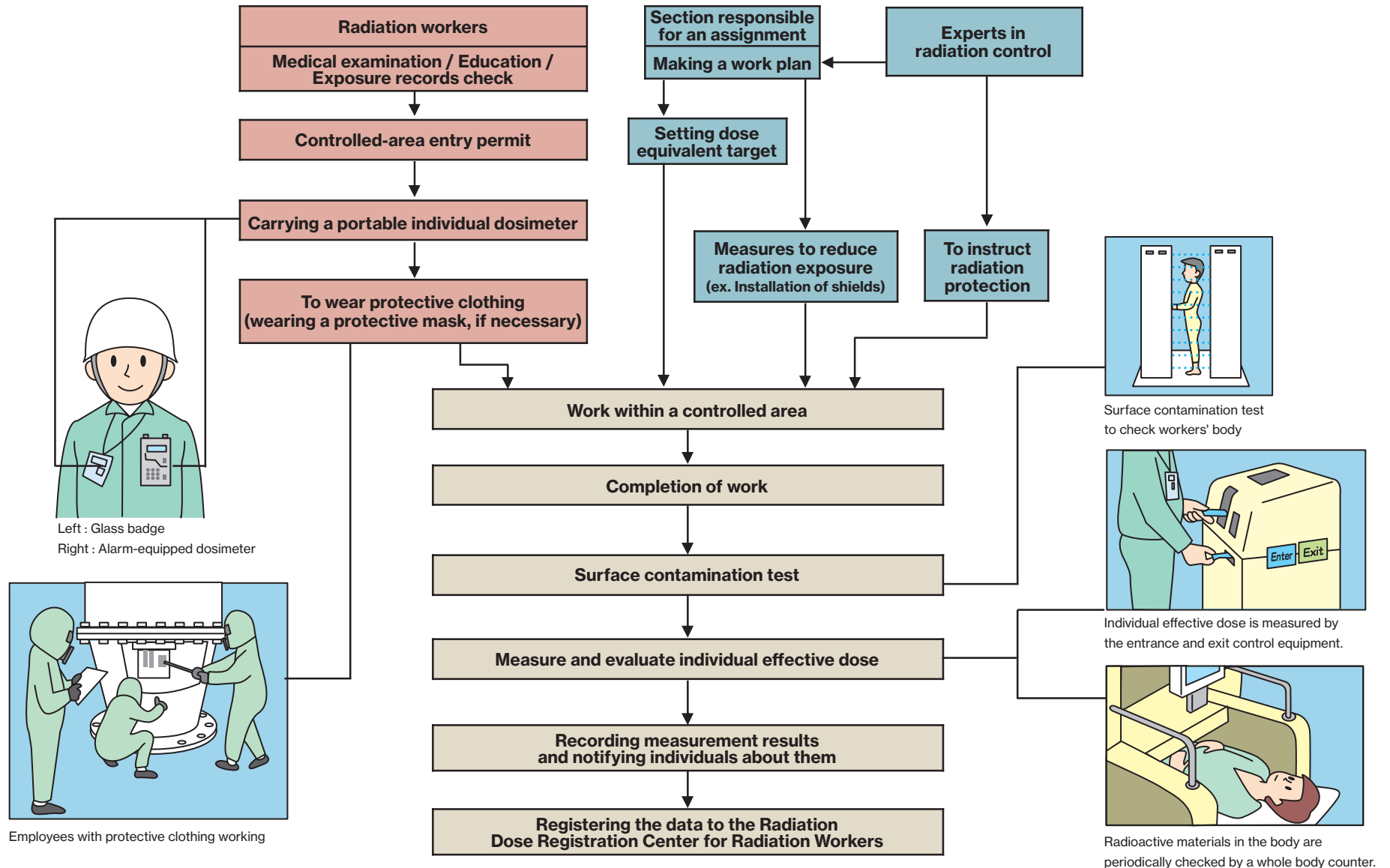
\* People who work with radiation at commercial nuclear power reactor facilities.

# Controls in a Nuclear Power Plant by Area



# Radiation Exposure Control for Radiation Workers

## Radiation Control Flow



# Radiation Dosage Limits

Category		Effective Dose Limit (Whole Body)	Equivalent Dose Limits (Tissues and Organs)
Radiation Workers	Normally	100mSv/5 years* <sup>1</sup> 50mSv/year* <sup>2</sup> Females 5mSv/3 months* <sup>3</sup> Pregnant women 1mSv (Internal exposure until birth)	Lens of eyes 100mSv/5 years* <sup>1</sup> , and 50mSv/year* <sup>2</sup> Skin 500mSv/year* <sup>2</sup> Pregnant women 2mSv (Abdominal surface until birth)
	Emergency* <sup>4</sup>	① 100mSv ② 250mSv	Lens of eyes 300mSv/year Skin 1Sv* <sup>5</sup>
General Public	Normally	1mSv/year* <sup>2</sup>	Lens of eyes 15mSv/year* <sup>2</sup> Skin 50mSv/year* <sup>2</sup>

(Note) The values in the table above are total doses of external and internal exposure. (They do not include naturally occurring radiation or exposure during medical procedures.)

\*1: Classification every 5 years after April 1, 2001

\*2: One year starting April 1

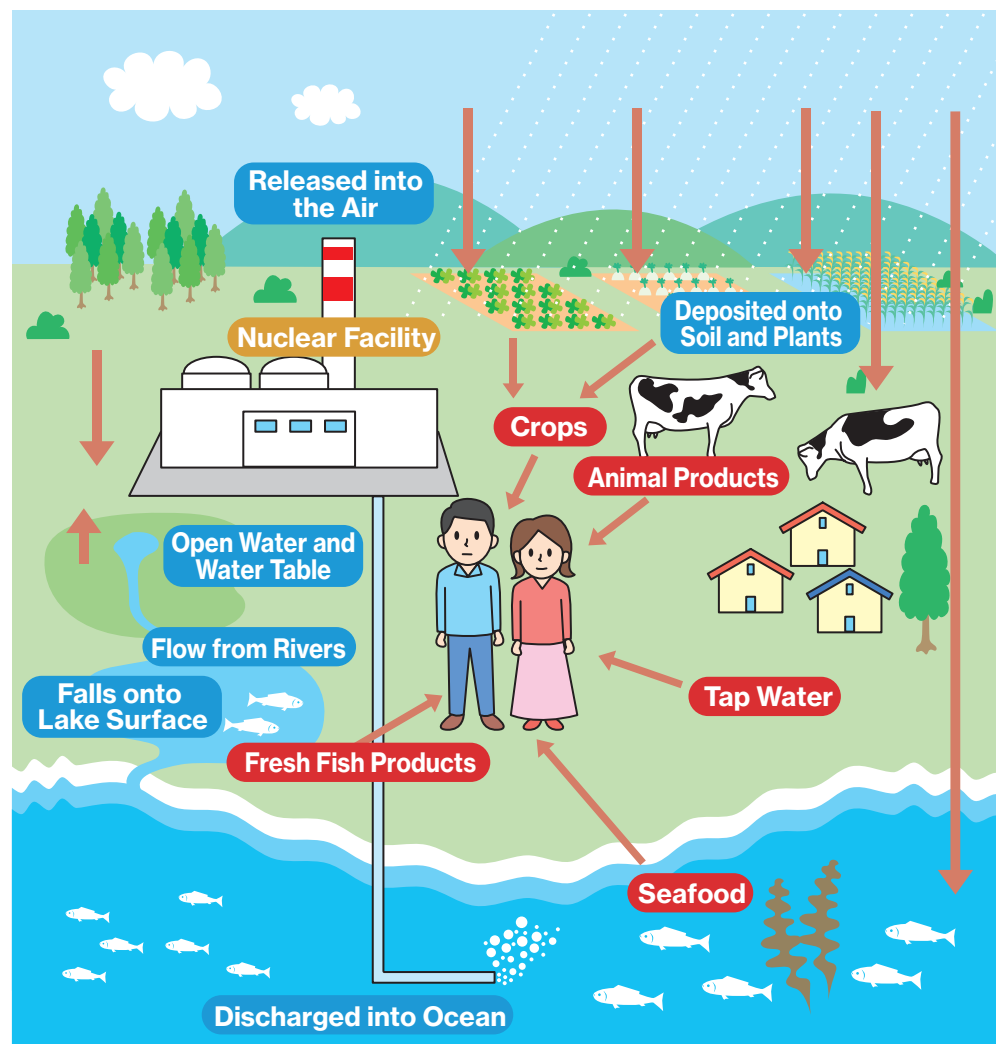
\*3: Every 3 months, starting on the 1st of April, July, October and January

\*4: • Engaging in emergency work on equipment and facilities, etc. targeted by the Act on Special Measures Concerning Nuclear Emergency Preparedness is restricted to workers in occupations that involve radiation exposure who have received information pertaining to radiation exposure in advance and have expressed their intention to take part, and who have received the necessary training.

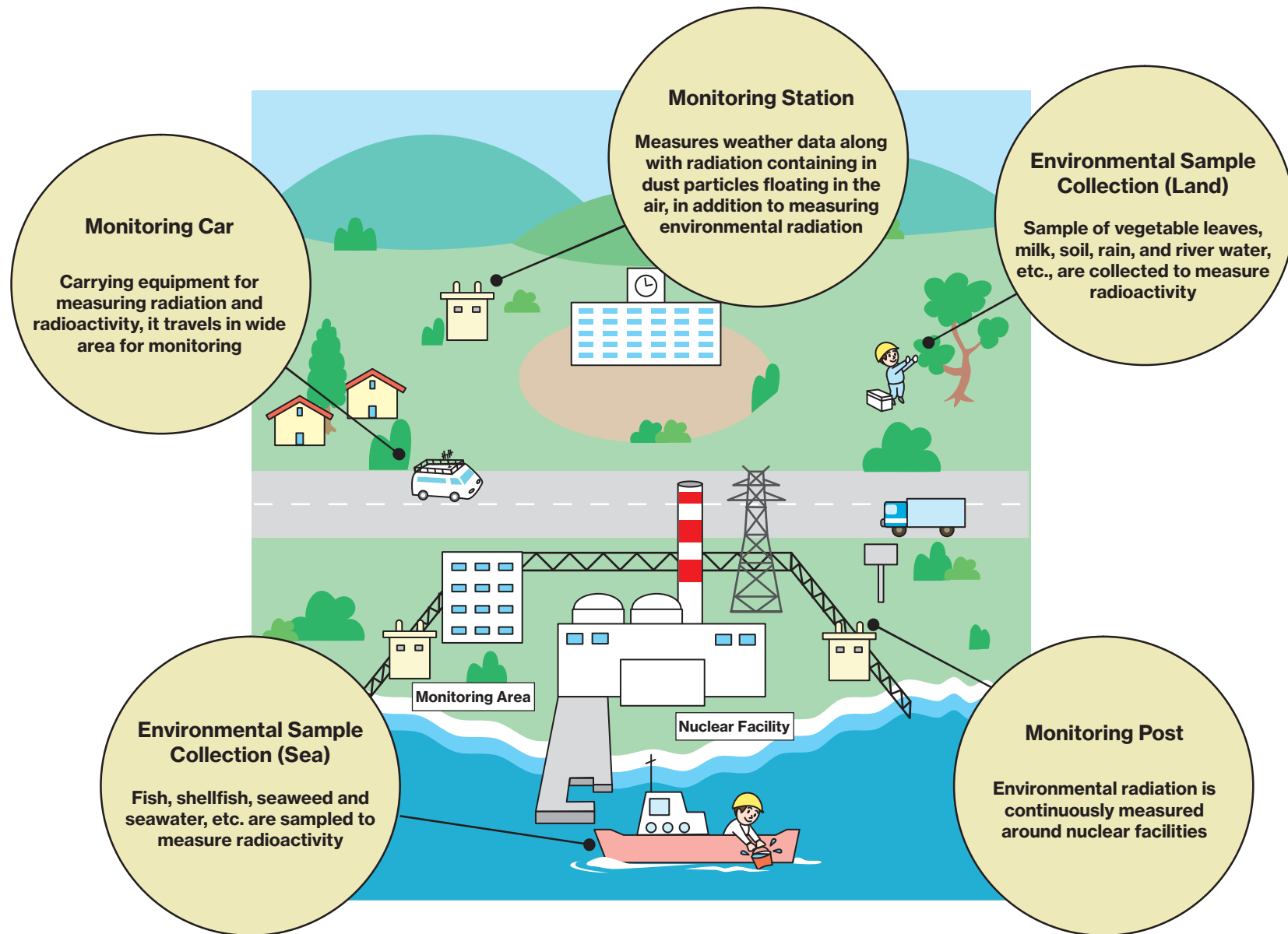
• Regulated limits on radiation exposure are governed in the two stages of ① the conventional effective dose of 100 mSv, and the addition of ② an effective dose of 250 mSv in the event of there being a high probability of radioactive materials being discharged outside of the site.

\*5: 1Sv (Sievert) = 1,000mSv (milli-Sieverts) = 1 million Sv (micro-Sieverts)

# Migration of Radioactive Materials in the Environment



# Environmental Radiation Monitoring Around Nuclear Facilities



# Monitoring of Radiation in the Environment (Example)



- Monitoring post and TLD post
- TLD\* post
- ★ Meteorological stations
- △ Seawater radiation monitoring
- Monitoring stations

**\*TLD : Thermoluminescence dosimeter**

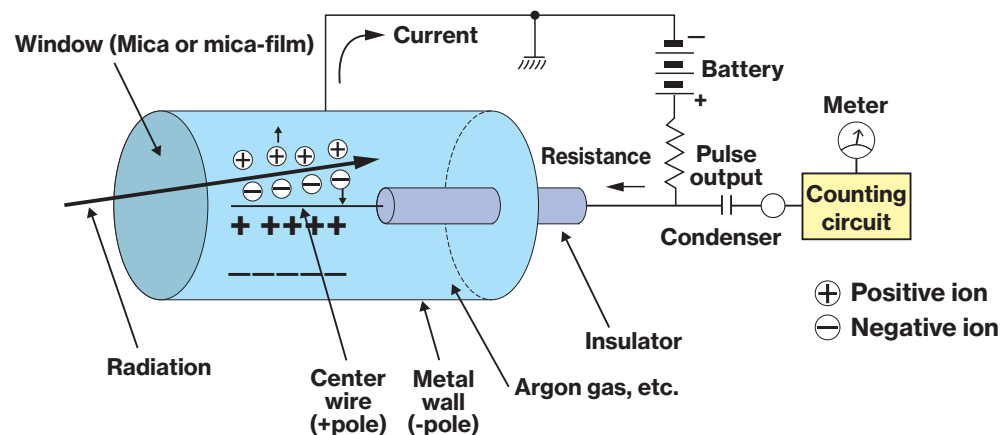
Some substances emit light if they are heated after being subject to radiation.

A TLD dosimeter uses this to measure radiation.

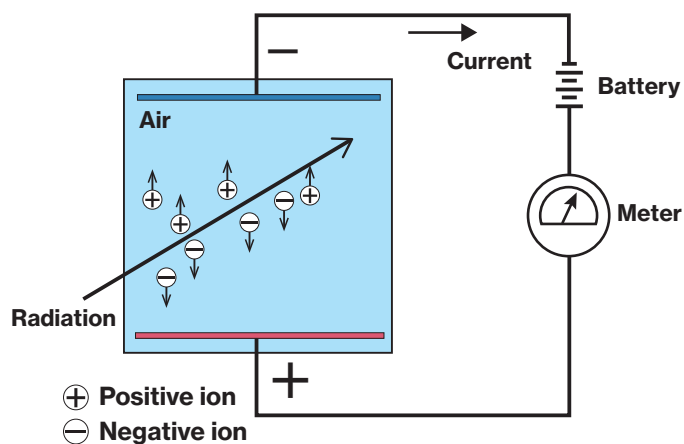


# Operating Principles of Radiation Measuring Instruments

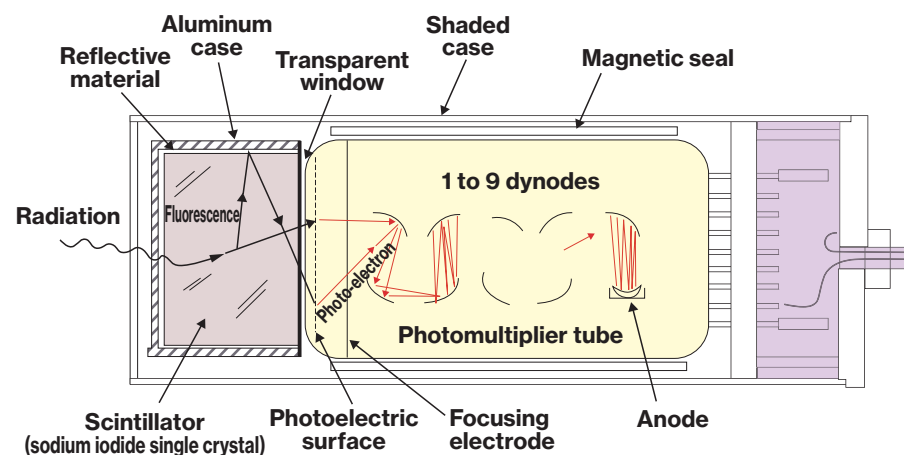
## Geiger Counter



## Ionization chamber



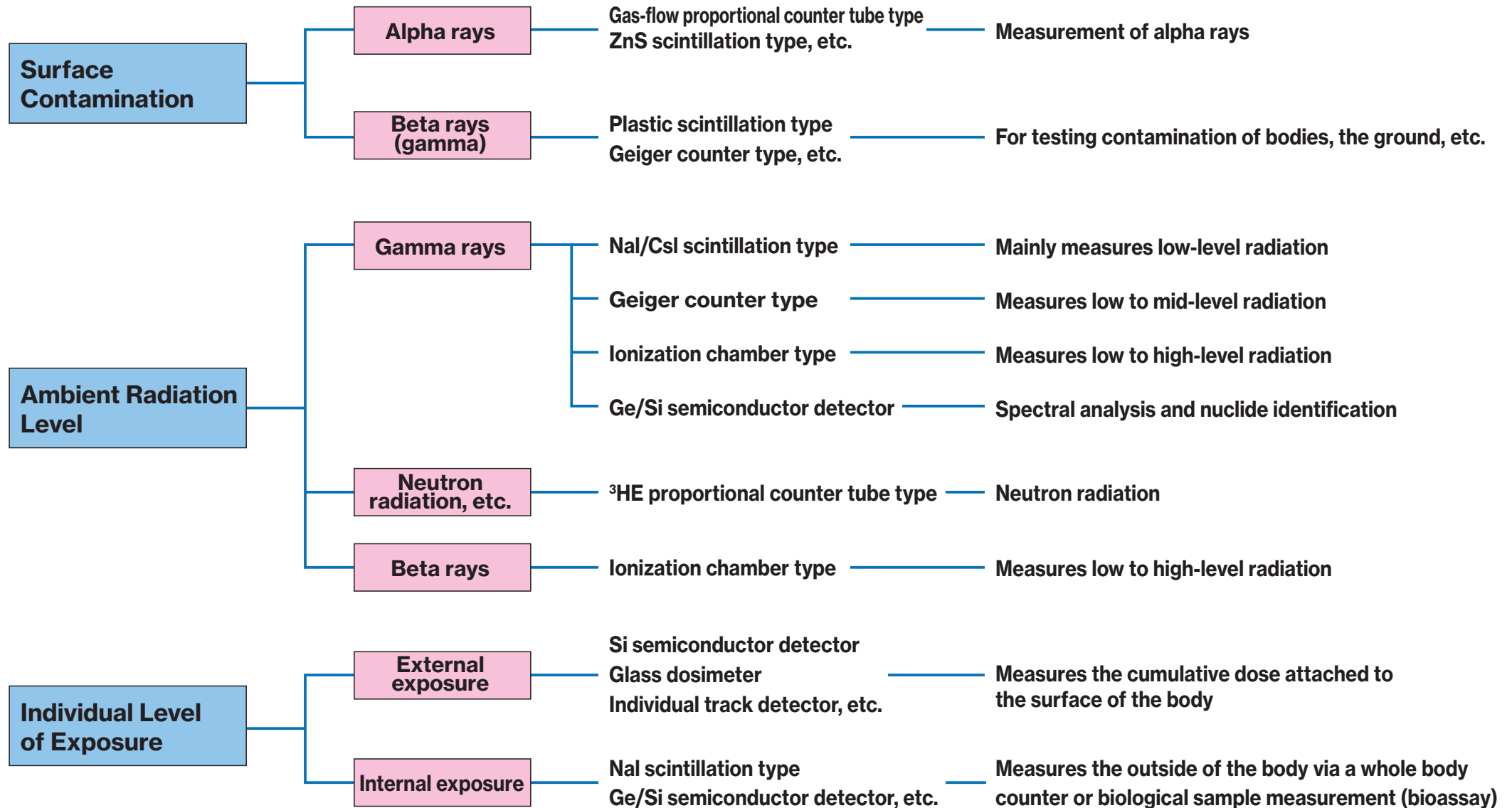
## Scintillation Detector



(Note) It is necessary to measure microcurrents in the range of  $10^{-9} \sim 10^{-14}$  with the ionization chamber.

When fluorescence strikes the photoelectric surface of the scintillation detector, electrons are kicked out and this is multiplied by the dynodes (multiplier electrodes), which allows a greater electrical signal to be obtained.

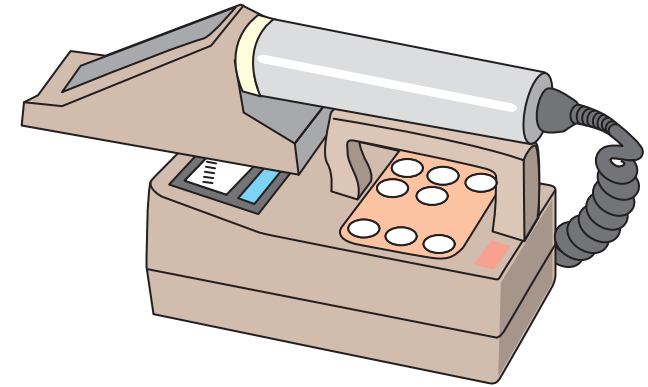
# Classifications of Radiation Measurements



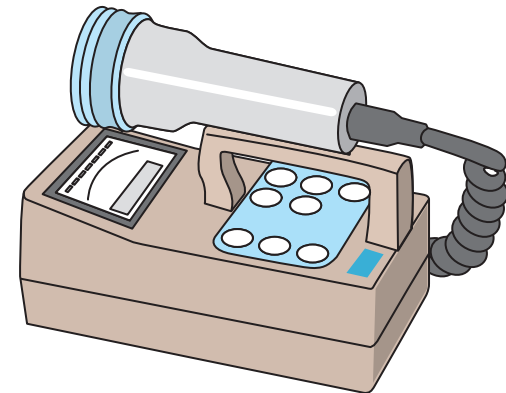
# Measuring Surface Contamination



**Radiation Screening**



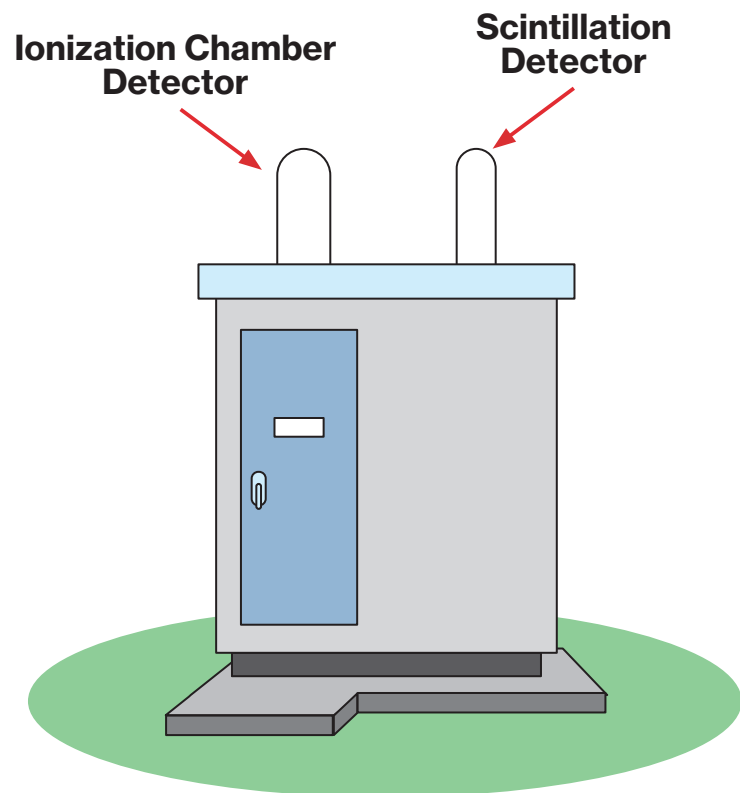
**ZnS scintillation type, etc.  
(measures alpha rays)**



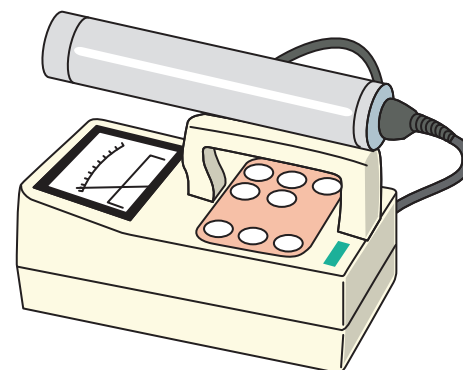
**Geiger counter type  
(measures beta rays)**

# Measuring Ambient Radiation Levels

## Monitoring Post



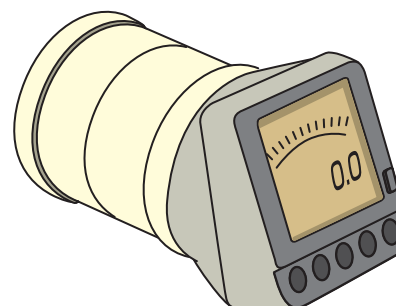
## Survey Meters



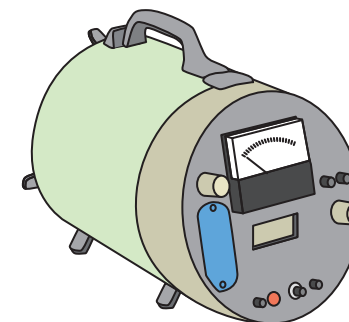
**NaI Scintillation type**  
(mainly measures low-levels)



**CsI Scintillation type, etc.**  
(mainly measures low-levels)



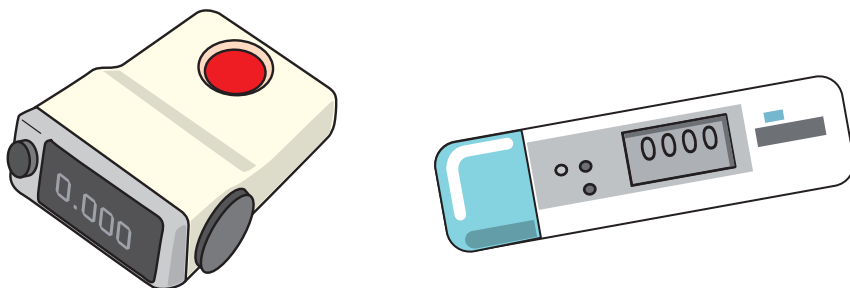
**Ionization Chamber**  
(measures low to high-level radiation)



**$^3\text{He}$  Counter Tube type**  
(measures neutron radiation)

# Measuring Individual Levels of Exposure Level of Exposure

## Measuring External Exposure

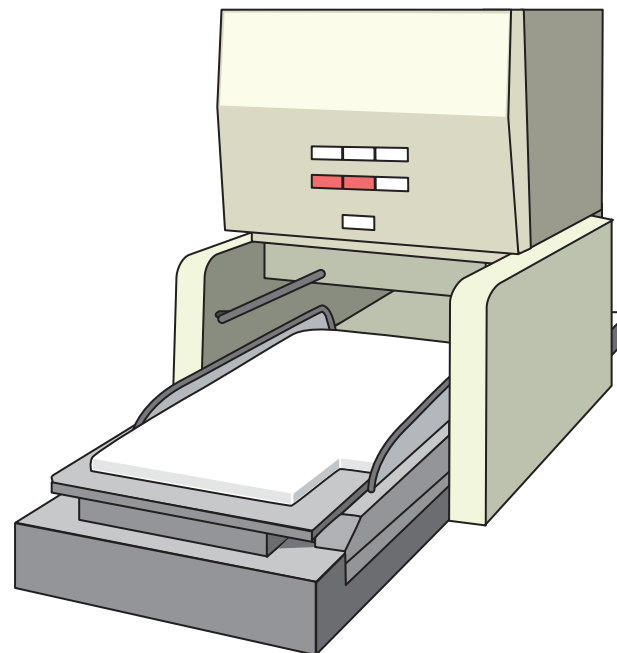


### Electronic Dosimeter (personal dosimeter)

[Example of Use]

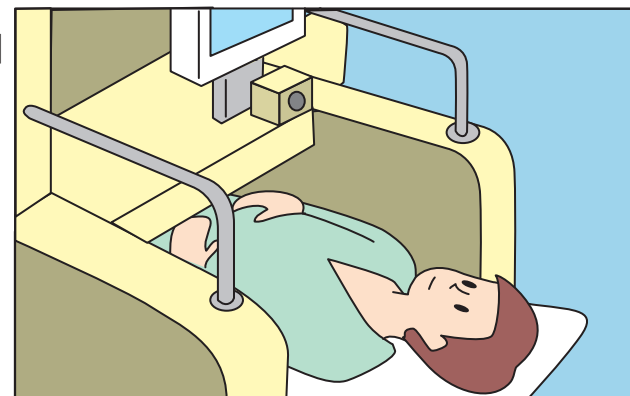


## Measuring Internal Exposure



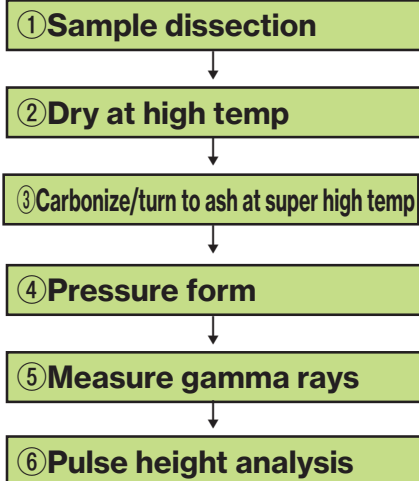
### Whole Body Counter

[Example of Use]

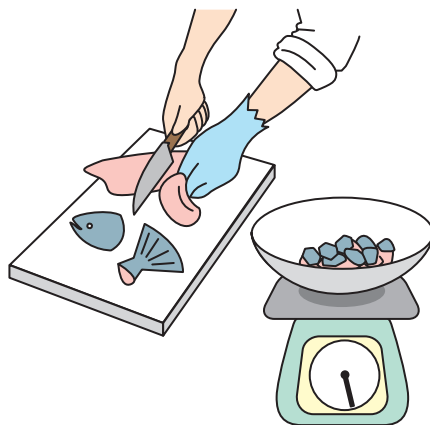


# Measuring Radioactivity Contained in Food

Workflow using a germanium semiconductor detector



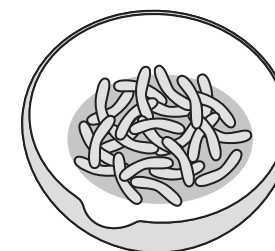
① Sample dissection



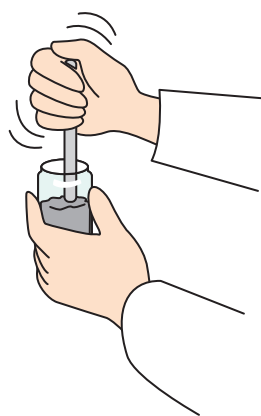
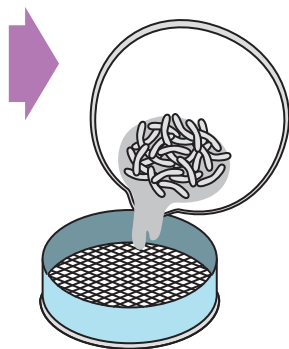
② Dry at high temp



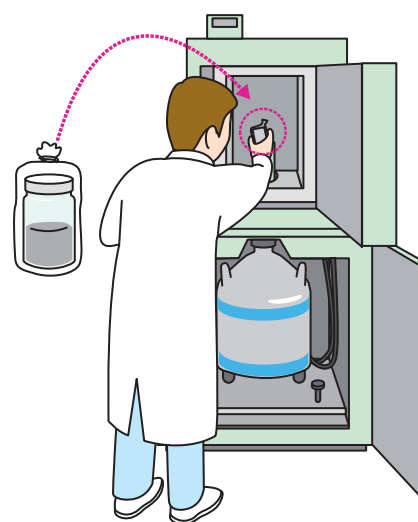
③ Carbonize/turn to ash at super high temp



④ Pressure form



⑤ Measure gamma rays



⑥ Pulse height analysis

